

12aiv

Uranium-238 ($^{238}_{92}\text{U}$)

12bi

Given:

Kinetic energy of electron (KE) $1.8 \times 10^{-16} \text{ J}$

Work function of metal $\phi = 1.20 \times 10^{-16}$

Planck's constant, $h = 6.6 \times 10^{-34} \text{ Js}$

$$E_{\text{photon}} = \phi + KE \\ = 1.20 \times 10^{-16} + 1.80 \times 10^{-16} = 3.00 \times 10^{-16} \text{ J}$$

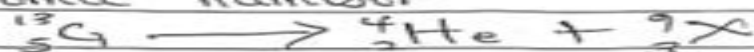
$$f = \frac{E_{\text{photon}}}{h} = \frac{3.00 \times 10^{-16}}{6.6 \times 10^{-34}} = \underline{\underline{4.545 \times 10^{17} \text{ Hz}}}$$

12bii

- (i) A cathode ray tube emits ~~out~~ electrons from the cathode
- (ii) A cathode ray tube produces a visible image on a phosphor-coated screen.

12c

I. In alpha decay, the nucleus loses 4 units of mass number and 2 units of atomic number



II. Daughter nucleus is ^9_3Li (Lithium)

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Patrick Vollmar



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Table of Contents Answer To Waec Physics Obj And Essay 2014 2015

1. Understanding the eBook Answer To Waec Physics Obj And Essay 2014 2015
 - The Rise of Digital Reading Answer To Waec Physics Obj And Essay 2014 2015
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer To Waec Physics Obj And Essay 2014 2015
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer To Waec Physics Obj And Essay 2014 2015
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer To Waec Physics Obj And Essay 2014 2015
 - Personalized Recommendations
 - Answer To Waec Physics Obj And Essay 2014 2015 User Reviews and Ratings
 - Answer To Waec Physics Obj And Essay 2014 2015 and Bestseller Lists
5. Accessing Answer To Waec Physics Obj And Essay 2014 2015 Free and Paid eBooks
 - Answer To Waec Physics Obj And Essay 2014 2015 Public Domain eBooks
 - Answer To Waec Physics Obj And Essay 2014 2015 eBook Subscription Services
 - Answer To Waec Physics Obj And Essay 2014 2015 Budget-Friendly Options
6. Navigating Answer To Waec Physics Obj And Essay 2014 2015 eBook Formats

- ePub, PDF, MOBI, and More
 - Answer To Waec Physics Obj And Essay 2014 2015 Compatibility with Devices
 - Answer To Waec Physics Obj And Essay 2014 2015 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer To Waec Physics Obj And Essay 2014 2015
 - Highlighting and Note-Taking Answer To Waec Physics Obj And Essay 2014 2015
 - Interactive Elements Answer To Waec Physics Obj And Essay 2014 2015
 8. Staying Engaged with Answer To Waec Physics Obj And Essay 2014 2015
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer To Waec Physics Obj And Essay 2014 2015
 9. Balancing eBooks and Physical Books Answer To Waec Physics Obj And Essay 2014 2015
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer To Waec Physics Obj And Essay 2014 2015
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Answer To Waec Physics Obj And Essay 2014 2015
 - Setting Reading Goals Answer To Waec Physics Obj And Essay 2014 2015
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Answer To Waec Physics Obj And Essay 2014 2015
 - Fact-Checking eBook Content of Answer To Waec Physics Obj And Essay 2014 2015
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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