Name:

Deute

Period:

Types of Chemical Reaction Worksheet

Balance the reactions 1 to 6 and indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented:

2 NaBr + Ca(OH)₂ à CaBr₂ + 2 NaOH

Reaction Type : single displacement

2. 2 NH₃+ H₂SO₄ à 3 (NH₄)₂SO₄

Reaction Type: synthesis

4 C₁H₀O + 27 O₂ à 20 CO₂ + 18 H₂O

Reaction Type: combustion

4. 3 Pb + 2 H₃PO₄ à 3 H₂ + Pb₃(PO₄)₂

Reaction Type : single displacement

Li₂N + 3 NH₄NO₂ à 3 LiNO₃ + (NH₄)₃N

Reaction Type: double displacement

HBr + 3 Al(OH)₃ à H₂O + 3 AlBr₃

Reaction Type: double displacement

Indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented in 7 to 20.

Na₃PO₄ + 3 KOH à 3 NaOH + K₃PO₄

Reaction Type double displacement

8. MgCl₂ + Li₂CO₃ à MgCO₃ + 2 LiCl

Reaction Type double displacement

9. C₆H₁₂ + 9 O₂ à 6 CO₂ + 6 H₂O

Reaction Type combustion

10. Pb + FeSO₄ à PbSO₄ + Fe

Reaction Type single displacement

11. CaCO₃ à CaO + CO₂

Reaction Type decomposition

12. P4 + 3 O2 à 2 P2O3

Reaction Type synthesis

13. 2 RbNO₃ + BeF₂ à Be(NO₃)₂ + 2 RbF

Reaction Type double displacement

14. 2 AgNO₃ + Cu à Cu(NO₃)₂ + 2 Ag

Reaction Type single displacement

15. C₃H₀O + 4 O₂ à 3 CO₂ + 3 H₂O

Reaction Type combustion

16. 2 C₅H₅ + Fe à Fe(C₅H₅)₂

Reaction Type synthesis

17. SeCl₅ + O₂ à SeO₂ + 3Cl₂

Reaction Type single displacement

18. 2 Mgl₂ + Mn(SO₃)₂ à 2 MgSO₃ + Mnl₄

Reaction Type double displacement

19. O3 à O + O2

Reaction Type decomposition

20. 2 NO2 à 2 O2 + N2

Reaction Type decomposition

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