

Chemistry, Part 1

Module 14 Notes



Big Question

How are metals used in chemical reactions? What products are produced?

Big Idea

Using chemical analysis, you can identify the products of a reaction between particles and other substances. A reaction between particles is used in chemical analysis to identify substances with chemical reactions.

Learning Objectives

After completing this module, students will be able to:

- Use the law of conservation of mass to balance chemical reactions.
- Use chemical reactions to identify substances.
- Compare different types of chemical reactions.
- Use the law of conservation of mass to identify substances.

Module 14 Big Idea

How are metals used in chemical reactions?

Fill in the blanks.

1. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
2. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
3. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.

Module 14 Big Idea

How are metals used in chemical reactions?

How are metals used in chemical reactions? What products are produced?

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Module 14 Big Idea

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Answers For Chemistry Module 14

Paul Collinson, Paul Collison



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