

# Anions, Cations, and Ionic Reactions

Hands-On Labs, Inc.

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## Lab Report Assistant

This document is not meant to be a substitute for a formal laboratory report. The Lab Report Assistant is simply a summary of the experiment's questions, diagrams if needed, and data tables that should be addressed in a formal lab report. The intent is to facilitate students' writing of lab reports by providing this information in an editable file which can be sent to an instructor.

## Exercise 1: Qualitative Anion Tests

Data Table 1. Anion Confirmation Tests.

| Chemical                 | Anion     | Observations: Addition of $\text{AgNO}_3$                   | Observations: Addition of $\text{HCl}$                                                                         |
|--------------------------|-----------|-------------------------------------------------------------|----------------------------------------------------------------------------------------------------------------|
| $\text{NaBr}$            | Bromide   | Turned milky white                                          | Turned yellowish<br>More of a gel                                                                              |
| $\text{Na}_2\text{CO}_3$ | Carbonate | Thick white glob<br>A little grainy                         | White sediment in a clear liquid<br>Bubbles up                                                                 |
| $\text{NaCl}$            | Chloride  | Milky white                                                 | White sediment visible                                                                                         |
| $\text{NaI}$             | Iodide    | Yellow glob immediately when<br>dropped in                  | Yellow, grainy/sediment, thick<br>looking                                                                      |
| $\text{Na}_2\text{SO}_4$ | Sulfate   | Thickened                                                   | Very cloudy/milky                                                                                              |
| $\text{Na}_2\text{S}$    | Sulfide   | Brown Globbs immediately when<br>dropped in<br>Still watery | Brown globbs thinned out into<br>brown pieces of sediment<br>Still watery (brownish) looks like<br>dirty water |
| Unknown<br>#103          | Carbonate | Thick white glob<br>Turned gel                              | Bubbly and cloudy<br>Very thick gel                                                                            |
| Unknown<br>#104          | Sulfide   | Brown globbs immediately when<br>dropped in<br>Still watery | Brown globbs thinned out into<br>brown pieces of sediment<br>Still watery (brownish) looks like<br>dirty water |

# Analysis Of Anions Cations Lab Answers

**Yi-Tong Ma**



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