

Question 1: Water at 20°C falls from a height of 854 m. If the whole energy is used in increasing the temperature, find out the final temperature. Specific heat of water is 4200 J/K kg .

Given that

$$T_1 = 20^{\circ}\text{C} = 20 + 273 = 293 \text{ K}$$

$$\text{height} = h = 854 \text{ m}$$

$$\text{Specific heat of water} = 4200 \text{ J K}^{-1} \text{ kg}^{-1}$$

$$\text{Final temperature} = T_2 = ?$$

$$\text{P.E} = mgh \quad \text{--- (1)}$$

Solution:

We know that

$$\Delta Q = cm \Delta T \quad \text{--- (2)}$$

Whole energy is used in increasing the temperature (Given)

$$\Delta Q = \text{P.E}$$

From (1) and (2)

$$cm \Delta T = mgh$$

$$c \Delta T = gh$$

$$c (T_2 - T_1) = gh$$

$$T_2 - T_1 = \frac{gh}{c}$$

$$T_2 = \frac{gh}{c} + T_1$$

$$= \frac{9.8 \times 854}{4200} + 293$$

$$= \frac{8369.2}{4200} + 293 = 1.99 + 293$$

$$T_2 = 294.99 \text{ K} = 295 - 273^{\circ}\text{C}$$

$$T_2 = 22^{\circ}\text{C}$$

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Thermodynamics of Energy Conversion and Transport Stanislaw Sieniutycz, Alexis De Vos, 2012-12-06 It is becoming more and more important to manage energy resources effectively to maximize their benefits while minimizing the negative environmental impacts Scientist and engineers are thus faced with the problem of optimizing complex systems subject to constraints from ecology economics and thermodynamics It is chiefly to the last that the present volume is addressed Nonequilibrium thermodynamic approaches such as finite time thermodynamics and Second Law analyses can provide realistic models and analyses that can be used to search for optimum ways to operate machines and processes Intended for physicists chemists and engineers this volume reviews the state of the art in the thermodynamics of energy conversion and transmission Using examples from solar thermal mechanical chemical and environmental engineering the book focuses on the use of thermodynamic criteria for optimizing energy conversion and transmission The first set of chapters focuses on solar energy conversion the second set discusses the transfer and conversion of chemical energy as in internal combustion engines or distillation columns a concluding set of chapters deals with geometric methods in thermodynamics

Theoretical Chemistry from the Standpoint of Avogadro's Rule & Thermodynamics Walther Nernst, 1895 **Ice and Refrigeration**, 1892 **Gravity and the Quantum** Jasjeet Singh Bagla, Sunu Engineer, 2017-02-23 This book provides a compilation of in depth articles and reviews on key topics within gravitation cosmology and related issues It is a celebratory volume dedicated to Prof Thanu Padmanabhan Paddy the renowned relativist and cosmologist from IUCAA India on the occasion of his 60th birthday The authors many of them leaders of their fields are all colleagues collaborators and former students of Paddy who have worked with him over a research career spanning more than four decades Paddy is a scientist of

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Biothermodynamics Part A Michael L. Johnson, Jo M. Holt, Gary K. Ackers, 2009-03-14 In the past several years there has been an explosion in the ability of biologists molecular biologists and biochemists to collect vast amounts of data on their systems This volume presents sophisticated methods for estimating the thermodynamic parameters of specific protein protein DNA and small molecule interactions The use of thermodynamics in biological research is used as an energy book keeping system While the structure and function of a molecule is important it is equally important to know what drives the energy force These methods look to answer What are the sources of energy that drive the function Which of the pathways are of biological significance As the base of macromolecular structures continues to expand through powerful techniques of molecular biology such as X ray crystal data and spectroscopy methods the importance of tested and reliable methods for answering these questions will continue to expand as well

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Kinetic Theory of Nonequilibrium Ensembles, Irreversible Thermodynamics,

and Generalized Hydrodynamics Byung Chan Eu, 2016-08-02 This book presents the fundamentals of irreversible thermodynamics for nonlinear transport processes in gases and liquids as well as for generalized hydrodynamics extending the classical hydrodynamics of Navier Stokes Fourier and Fick Together with its companion volume on relativistic theories it provides a comprehensive picture of the kinetic theory formulated from the viewpoint of nonequilibrium ensembles in both nonrelativistic and in Vol 2 relativistic contexts Theories of macroscopic irreversible processes must strictly conform to the thermodynamic laws at every step and in all approximations that enter their derivation from the mechanical principles Upholding this as the inviolable tenet the author develops theories of irreversible transport processes in fluids gases or liquids on the basis of irreversible kinetic equations satisfying the H theorem They apply regardless of whether the processes are near to or far removed from equilibrium or whether they are linear or nonlinear with respect to macroscopic fluxes or thermodynamic forces Both irreversible Boltzmann and generalized Boltzmann equations are used for deriving theories of irreversible transport equations and generalized hydrodynamic equations which rigorously conform to the tenet All observables described by the so formulated theories therefore also strictly obey the tenet **Chemical Thermodynamics**

For Metals And Materials (With Cd-rom For Computer-aided Learning) Hae-geon Lee, 1999-10-13 A number of thermodynamic books claiming to be original in both presentation and approach have been published However thermodynamics is still a confusing subject for uninitiated students and an easy to forget one for graduate engineers In order to solve these problems this computer aided learning package textbook and CD ROM takes a new approach This package is unique and beneficial in that it simulates a classroom lecture it actually writes important equations and concepts on a virtual board underlines draws circles places ticks to emphasise important points draws arrows to indicate relationships uses colours for visual effect erases some parts to write new lines and even repeats some parts of the lesson to stress their importance This realistic simulation is made possible by the employment of the multimedia capabilities of the modern day computer Readers are not just passively presented with thermodynamics they can also interactively select and repeat any particular topic of interest as many times as they want This flexibility allows readers to choose their own pace of presentation This complementary set is in many important respects better than the books that are currently available on the subject

Analytical Thermodynamics Dongqing Li, 2024-12-07 This second edition presents an enriched and expanded exploration of the fundamental principles of thermodynamics tailored for graduate level studies Drawing on over three decades of academic teaching experience the author has refined the content making it more accessible and comprehensive Chapter 1 has been restructured for clarity delineating Legendre Transformation and Thermodynamic Potentials into separate sections while the treatment of Chemical Potentials has been significantly augmented encompassing two component ideal gas mixtures and a re derivation of chemical potentials for dilute solutions Additionally the section on thermodynamic stability now boasts enhanced explanations and illustrative figures Chapter 2 introduces a groundbreaking section Electrolyte

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