

A chemical bond - forms when 2 or more atoms rearrange _____
_____ to increase _____.

ionic bond - forms when valence _____ and _____
from one atom to another.

cation - atom _____ electrons to become _____ charged.

anion - atom _____ electrons to become _____ charged.

In ionic compounds the ions are arranged in a _____
_____ forces hold the ions together.

Properties of ionic compounds:

- High _____ and _____ points
- _____ - not easily _____
- _____ electricity when _____ or
_____ because the ions are free to _____.

covalent bond - _____ and _____, forming
_____.

Covalent compounds have _____ forces holding the
_____ together.

Properties of covalent compounds:

- lower _____ and _____ points
- Many covalent compounds are _____ liquids or gases.
- _____ - easier to _____
- are not _____ of electricity

electronegativity - property that tells how strong an atom's _____
is for _____.

Since oxygen has a _____ electronegativity than hydrogen, oxygen
holds onto shared electrons _____, giving the oxygen a _____
negative charge and the hydrogen a partial _____ charge.

Answers For Note Taking Guide Episode 501

**Anna Lorraine Guthrie, Bertha
Tannehill, Neltje Marie Tannehill
Shimer**



Answers For Note Taking Guide Episode 501:

Unveiling the Magic of Words: A Report on "**Answers For Note Taking Guide Episode 501**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Answers For Note Taking Guide Episode 501**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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