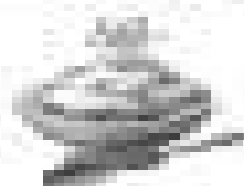


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine which elements are present in a sample of an element by using the emission spectrum.

Part I: Finding Evidence for Discrete Energy Levels

When an element is excited, it emits light. The light is composed of discrete energy levels. The energy levels are discrete, meaning that the energy levels are not continuous. The energy levels are discrete, meaning that the energy levels are not continuous. The energy levels are discrete, meaning that the energy levels are not continuous.

Objective: To determine which elements are present in a sample of an element by using the emission spectrum.

Using the emission spectrum to determine which elements are present in a sample of an element by using the emission spectrum.

Procedure: The student will use the emission spectrum to determine which elements are present in a sample of an element by using the emission spectrum. The student will use the emission spectrum to determine which elements are present in a sample of an element by using the emission spectrum.

Using the emission spectrum to determine which elements are present in a sample of an element by using the emission spectrum.

Conclusion: The student will use the emission spectrum to determine which elements are present in a sample of an element by using the emission spectrum.

Using the emission spectrum to determine which elements are present in a sample of an element by using the emission spectrum.

Answers Lesson 11 Atomic Pudding

C Cleary



Answers Lesson 11 Atomic Pudding:

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