



# Nuclear Radiation Worksheet

a. What determines whether an atom is stable or unstable?

It's overall size and a difference between proper it and prentom it

b. How does an unstable atom gain stability?

It undergoes rediseasing decay

c. What part of an atom undergoes change during radioactive decay?

The Mudeogi

2. Which of the human senses can detect radiation?

ich of the human senses can detect radiation?
Our eyes can dutect colors, but we have no senses to detect alpha, Beta, or gamen.

Explain the difference between an isotope and a radioactive isotope.
 Δε τριέτες λέο α αλέστες έτες ρομένες (καλούς καλούς το league orchétic and has a l'Accord at 4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	of sign 5 persons	Power (4)	Shrand by boas of poor
Beta	O No Rest 10 crem at N	Dienetive (-)	compact by sheets shall be think may zone
Gamma	O No our	Neubol	Should be small inches of food or feet

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	- H	-2
Beta	No change	+1
Gamma	No element	No charge

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.

Example:  ${}^{218}_{84}Po \rightarrow {}^{4}_{2}He + {}^{214}_{82}Pb$ 

- Complete the following nuclear equations:
- 239 Bi → 4He + 235 T/ b. <sup>239</sup><sub>93</sub> Np → <sup>239</sup><sub>94</sub> Pu + □

# **Answer Key For Chemistry Radioactive Decay Worksheet**

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