

Chapter 19

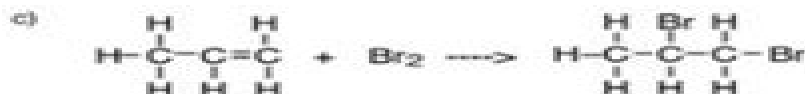
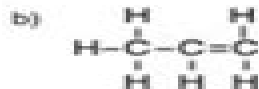
1. a) Contains as many hydrogen atoms as possible for the given number of carbons. Only carbon-carbon single bonds.

b) (i) $C_{11}H_{24}$

(ii) liquid



2. a) a carbon-carbon double bond



(Home schooling parents: It doesn't matter what position you draw the bromine in on the right-hand carbon - N, S or E (with hydrogens in the other two positions). All that matters is that one bromine becomes attached to each of the carbons either side of the original double bond.)



(ii) In this (substitution) reaction, one of the hydrogens has been replaced by a bromine during the reaction. In the previous (addition) reaction, nothing was lost when the two molecules combined together.

3. a) Ethene is passed with steam at $300^\circ C$ and 60 - 70 atmospheres over a phosphoric acid catalyst.

b) e.g. two of: continuous flow process, therefore faster; rapid reaction; purer product.

c) uses renewal resources; gentle conditions.

4. Dissolve the sugar in water in a conical flask (or similar) and add yeast. You now need some means of keeping the air out while allowing carbon dioxide to escape. You could use a commercial home-winemaking airlock of some sort, but a plug of cotton wool works just as well. Leave the mixture in the warm for a few days until bubbling stops.

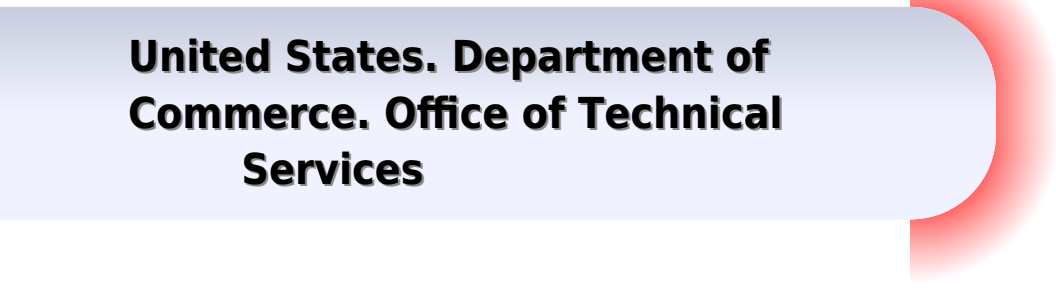


Allow the yeast to settle, and pour off the solution so that it can be fractionally distilled, collecting the fraction which boils at $78^\circ C$. (See the diagram for the fractional distillation of an alcohol/water mixture in Chapter 11 of the book.)

continued . . .

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Table of Contents Answer Key To Pearson Catalyst Lab Manual

1. Understanding the eBook Answer Key To Pearson Catalyst Lab Manual
 - The Rise of Digital Reading Answer Key To Pearson Catalyst Lab Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Pearson Catalyst Lab Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Pearson Catalyst Lab Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Pearson Catalyst Lab Manual
 - Personalized Recommendations
 - Answer Key To Pearson Catalyst Lab Manual User Reviews and Ratings
 - Answer Key To Pearson Catalyst Lab Manual and Bestseller Lists
5. Accessing Answer Key To Pearson Catalyst Lab Manual Free and Paid eBooks
 - Answer Key To Pearson Catalyst Lab Manual Public Domain eBooks
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 - Answer Key To Pearson Catalyst Lab Manual Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Answer Key To Pearson Catalyst Lab Manual Compatibility with Devices
 - Answer Key To Pearson Catalyst Lab Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To Pearson Catalyst Lab Manual
 - Highlighting and Note-Taking Answer Key To Pearson Catalyst Lab Manual
 - Interactive Elements Answer Key To Pearson Catalyst Lab Manual

8. Staying Engaged with Answer Key To Pearson Catalyst Lab Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Pearson Catalyst Lab Manual
9. Balancing eBooks and Physical Books Answer Key To Pearson Catalyst Lab Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Pearson Catalyst Lab Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key To Pearson Catalyst Lab Manual
 - Setting Reading Goals Answer Key To Pearson Catalyst Lab Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key To Pearson Catalyst Lab Manual
 - Fact-Checking eBook Content of Answer Key To Pearson Catalyst Lab Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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