



Nuclear Radiation Worksheet

a. What determines whether an atom is stable or unstable?

It's overall size and a difference between proper it and prentom it

b. How does an unstable atom gain stability?

It undergoes redisastive decay

c. What part of an atom undergoes change during radioactive decay?

The Mucheus !

Which of the human senses can detect radiation?

ich of the human senses can detect radiation?
Our eyes can detect colors, but we have no senses to detect alpha, between patriotion

Explain the difference between an isotope and a radioactive isotope.
 Isotope has a defend that payou feeders. Radiodox 15 league modelle and has a defend at 4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	The state of	Positive C+1	Storped by files in proper
Beta	O No Ress (Garren L.)	Directive (+)	compathy duch of his full or this impact
Gamma	B. No mark	Nambral	Should be sever when it had or fact

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	-4	-2
Beta	No change	41
Gamma	No deserve	No chance

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.

Example: ${}^{218}_{84}Po \rightarrow {}^{4}_{2}He + {}^{214}_{82}Pb$

a.
$$^{210}Pb \rightarrow ^{610}_{83}Bi + ^{6}_{-1}e^{-\beta}$$

b.
$$^{230}Th \rightarrow ^{17}Ra + ^{4}He$$

- 7. Complete the following nuclear equations:
- $^{239}_{83}Bi \rightarrow ^{4}_{2}He + ^{235}_{21}\mathcal{T}$ b. 239 Np → 239 Pu + -1

Answer Key To Nuclear Radiation

L Reisser

Answer Key To Nuclear Radiation:

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