

Lab objective:

Materials:

- | | |
|------------------------------------|-----------------------|
| - Vinegar (5% acetic acid) | - Graduated cylinders |
| - Baking soda (sodium bicarbonate) | - Ziploc bag |
| - Electronic scale | - Calculator |

Procedure:

1. Balance the chemical reaction of baking soda and vinegar:



2. Get a Ziploc bag
3. Record:
temperature: _____ Pressure: _____ in the room.
4. Determine the volume of your Ziploc bag in liters: _____
5. Calculate the mass of baking soda (sodium bicarbonate) required to completely fill your airbag with CO_2 .

6. Calculate the volume vinegar (5% acetic acid) required to completely fill your airbag. Vinegar has an acetic acid, $\text{HC}_2\text{H}_3\text{O}_2$, concentration of 0.833 mol / L.

Airbag Stoichiometry Lab Work And Answers

Thomas Bond,Chris Hughes



Airbag Stoichiometry Lab Work And Answers:

The Enigmatic Realm of **Airbag Stoichiometry Lab Work And Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Airbag Stoichiometry Lab Work And Answers** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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