# Metallic bonding

- What happens when a metal bonds with another metal?
- Both metal atoms give up their valence electrons.

 This forms a "sea of electrons" that fills the space between the metal cations. The attraction of the positively charged metal cations to the negative valence electrons is what

holds the metal together.



## **Answers For Ionic Metallic Bonding**

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This book delves into Answers For Ionic Metallic Bonding. Answers For Ionic Metallic Bonding is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Answers For Ionic Metallic Bonding, encompassing both the fundamentals and more intricate discussions.

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  - ∘ Chapter 5: Conclusion
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