

10-1

Practice (continued)

Form G

Areas of Parallelograms and Triangles

18. A company wants to paint its logo on the side of a building. The entire area needs to be covered with a primer. The two triangular areas will be painted red, the rectangle containing the company's name will be white, and the rest of the parallelogram will be yellow.

- Find the area for each different color.
- Find the area that must be painted with primer.



Handwritten notes on the right side of the page:

- Area of parallelogram = 1/2 * base * height = 1/2 * 10 * 8 = 40 sq ft
- Area of rectangle = length * width = 4 * 4 = 16 sq ft
- Area of two triangles = 40 - 16 = 24 sq ft

19. A scale drawing of the side view of a house is shown at the right. Find the total area (in square inches) of the side of the house in the drawing.



20. **Open-Ended** Using graph paper, draw a figure with area 42 units² made up of a parallelogram and a triangle.

Coordinate Geometry Find the area of a polygon with the given vertices.

- $A(2, 2), B(5, 2), C(3, -1), D(0, -1)$
- $A(1, 4), B(-2, -2), C(-7, -2), D(-4, 4)$
- $A(5, -3), B(-1, -3), C(-1, 2), D(5, 6)$
- $A(5, 0), B(5, 8), C(-1, 7), D(-1, -6)$

Find the area of each figure.



28. **Reasoning** A parallelogram has a height of 6 units and a corresponding base of 7 units. What is the area of each triangle formed when one diagonal of the parallelogram is drawn? What is the area of each small triangle formed when two diagonals are drawn?
29. A parallelogram has sides 24 m and 5 m. The height corresponding to a 24-m base is 4 m. What is the height corresponding to a 5-m base?

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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that
volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the
experiment looked at the relationship between temperature and pressure

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266 section 01 introduction in this lab we observed the relationship between pressure and temperature

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the
syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the

syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

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web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts