

Lab 10: Calculations with a Chemical Structure

Objectives:

1. To derive the empirical formula of a pure substance with percent composition (molecular formula) information.
2. To calculate the number of moles of a solid from weighing information in the lab.
3. To determine the molar mass of a solid.
4. To determine the theoretical amount of product that would be produced.
5. To compare the theoretical amount to the actual amount of substance formed and calculate the percent yield.

Reaction Information:

Reaction of Sodium Chloride and Silver Nitrate: $NaCl + AgNO_3 \rightarrow AgCl + NaNO_3$

Reaction of Silver Nitrate and Sodium Chloride: $AgNO_3 + NaCl \rightarrow AgCl + NaNO_3$

Reaction of Silver Nitrate and Sodium Chloride: $AgNO_3 + NaCl \rightarrow AgCl + NaNO_3$

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Reaction of Silver Nitrate and Sodium Chloride: $AgNO_3 + NaCl \rightarrow AgCl + NaNO_3$

Procedure:

Reaction and Calculations:

1. Reaction:

$$\text{moles of sodium chloride} = \frac{\text{mass of NaCl}}{\text{molar mass of NaCl}} = \frac{0.50 \text{ g}}{58.44 \text{ g/mol}} = 0.00855 \text{ mol}$$

$$\text{moles of silver nitrate} = \frac{\text{mass of AgNO}_3}{\text{molar mass of AgNO}_3} = \frac{0.50 \text{ g}}{169.87 \text{ g/mol}} = 0.00294 \text{ mol}$$

$$\text{moles of silver chloride produced} = \text{moles of AgNO}_3 = 0.00294 \text{ mol}$$

$$\text{mass of silver chloride produced} = \text{moles of AgCl} \times \text{molar mass of AgCl} = 0.00294 \text{ mol} \times 143.32 \text{ g/mol} = 0.421 \text{ g}$$

2. Reaction: $NaCl + AgNO_3 \rightarrow AgCl + NaNO_3$

3. Reaction:

$$\text{mass of AgCl} = \text{moles of AgCl} \times \text{molar mass of AgCl} = 0.00294 \text{ mol} \times 143.32 \text{ g/mol} = 0.421 \text{ g}$$

$$\text{mass of AgCl} = \text{moles of AgCl} \times \text{molar mass of AgCl} = 0.00294 \text{ mol} \times 143.32 \text{ g/mol} = 0.421 \text{ g}$$

$$\text{mass of AgCl} = \text{moles of AgCl} \times \text{molar mass of AgCl} = 0.00294 \text{ mol} \times 143.32 \text{ g/mol} = 0.421 \text{ g}$$

Answers For Lab 6b Heath Chemistry

Arturo Cuomo



Answers For Lab 6b Heath Chemistry:

Embracing the Song of Appearance: An Psychological Symphony within **Answers For Lab 6b Heath Chemistry**

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