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web nov 20 2020 in this video we ll balance the equation $\text{mg} + \text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{no}_2 + \text{h}_2\text{o}$ and provide the correct coefficients for each compound to balance $\text{mg} + \text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{no}_2 + \text{h}_2\text{o}$ you ll need to be

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web magnesium hydroxide nitric acid balanced total equation $\text{mg}(\text{oh})_2 + \text{aq} + 2 \text{hno}_3 \rightarrow \text{aq} + 2 \text{h}_2\text{o} + \text{mg}(\text{no}_3)_2$ total ionic equation

$\text{mg}^{2+} + 2 \text{oh}^- + 2 \text{h}^+ + 2 \text{no}_3^- \rightarrow \text{mg}^{2+} + 2 \text{no}_3^- + 2 \text{h}_2\text{o}$ net ionic equation $2 \text{oh}^- + 2 \text{h}^+ \rightarrow 2 \text{h}_2\text{o}$ sulfuric acid and lithium hydroxide

balanced total equation $\text{h}_2\text{so}_4 + \text{aq} + 2 \text{lioh} \rightarrow \text{aq} + 2 \text{h}_2\text{o}$