

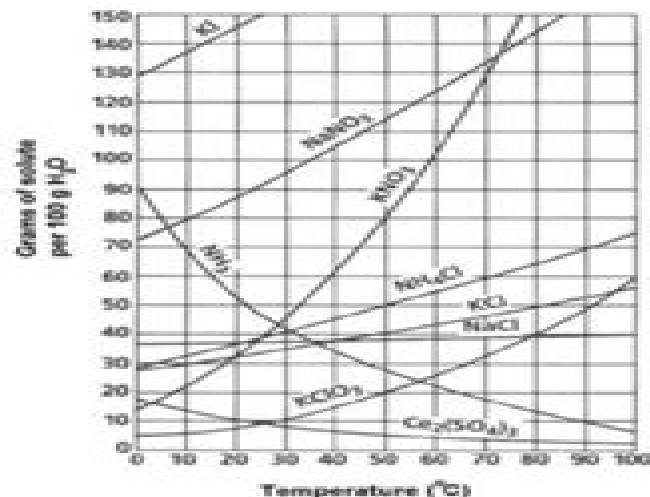
Solubility worksheet

Write the correct term (or terms) in the space provided.

1. As temperature increases, the solubility of gases in liquids generally _____.
2. When a solute dissolves and recrystallizes at the same rate, the solution is at _____.
3. A substance that does not dissolve in a polar solvent is probably _____.
4. To carry an electric current, a solution must contain _____.
5. A solution that contains more dissolved solute than a saturated solution contains under the same conditions is called a(n) _____.
6. The solubility of $\text{CO}_2(\text{g})$ would _____ with increasing solvent temperature.

Based on the graph answer the following questions:

1. Which compound is LEAST soluble at 10°C ? _____
2. How many grams of KCl can be dissolved in 100g of water at 80°C ? _____
3. How many grams of NaCl can be dissolved in 100g of water at 90°C ? _____
4. At 40°C , how much KNO_3 can be dissolved in 100g of water? _____
5. A saturated solution of KClO_3 is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C , how many grams of precipitate are formed? _____



What are 3 factors that affect solubility?

- a) _____
- b) _____
- c) _____

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