- 2. Assertion: Chips and other snacks become rancid due to oxidation.
  - Chips and other snacks become to the snacks they are packed in cartes Reason:

#### Answers 1.3

- (b) It is due to formation of Ag<sub>a</sub>S.
  - (d) He or N<sub>2</sub> because these are inert gases.
  - 3. (b) CuCO<sub>3</sub>.Ču(OH)<sub>2</sub> is green.
  - (d) Fe<sub>2</sub>O<sub>3</sub>xH<sub>2</sub>O is the chemical formula of reddish brown rust.
  - 5. (c) BHT (Butylated hydroxy toluene)
- 1. (a) Both 'A' and 'R' are true and 'R' is correct explanation of the assertion.
  - (c) 'A' is true but 'R' is false. Foods are packed in inert gases like nitrogen to prevent from rancidity.

# CASE STUDY QUESTIONS

 A student took HCl in a conical flask and was placed on a white tile with cross mark. When seen through the flask the mark was visible. On adding some sodium thiosulphate in the flask the cross mark disappeared when seen through the flask. The student observed some reaction as the products formed were different than the reactants. Both the reactants were transparent.



- (i) Name the type of reaction seen in the set up.
- (a) Combustion (b) Combination (c) Precipitation (ii) Why did the cross mark disappear?
  - (a) Decomposition
  - (a) The solution formed in the flask does not allow the light to pass through it to see the cross mark
  - (b) The sulphur is a colloid in this reaction, staying in suspension and eventually blocking the light from reaching the solution
  - (c) The solution transforms from being colourless to being milky and entirely opaque (d) All of the above
- (iii) Select the correct reaction for the above chemical reaction
  - $(a) \ \operatorname{Na_2S_2O_3}\left(\operatorname{aq}\right) + 2\operatorname{HCl}\left(\operatorname{aq}\right) \longrightarrow 2\operatorname{NaCl}\left(\operatorname{aq}\right) + \operatorname{H_2O(l)} + \operatorname{SO_2}\left(\operatorname{g}\right) + \operatorname{S(s)}$
  - (b)  $S_2O_3^{-2}$  (aq) + 2H\* (aq)  $\longrightarrow$  H<sub>2</sub>O (l) +  $SO_2$  (g) + S(s)
  - (c)  $Na_2S_2O_3$  (aq) + 2HCl (aq)  $\longrightarrow$  2NaCl (aq) +  $SO_2$  (g) + S(8)
  - (d) None of the above
- (iv) Name the resultant insoluble substance formed in the flask
  - (a) Copper(II) iodide (c) Sulphur
- (v) What type of reaction occurs in the flask?
  - (a) Displacement reaction
- (c) Photo-chemical decomposition Ans. (i) (c)(ii) (cl)(iii) (a)
- (b) Potassium hydroxide
- (d) Sodium Thiosulphate
- (b) Double displacement reaction
- (d) Thermal decomposition
- (iv) (c)(v) (b)

# **2014 Msce Questions Physical Science**

**Thomas Griffiths** 

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in christianity the holy spirit is seen as the third person of the trinity along with the father god and the son jesus 1 the acts of the apostles or the book of acts or simply acts is the fifth book of the christian new testament following immediately after jesus s crucifixion the book tells of the founding of the christian

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luke acts is an attempt to answer a theological problem namely how the messiah of the jews came to have an overwhelmingly non jewish church the answer it provides is that the message of christ was sent to the gentiles because as a whole jews rejected it 3