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AP CHEMISTRY MULTIPLE CHOICE QUESTIONS

Unit 8 – Introduction to Equilibrium



The equilibrium constant for the gas phase reaction above is 95 at 25°C. What is the value of the equilibrium constant for the following reaction at 25°C?



Reversed
+ Doubled

(A) 95²

(B) 95

(C) (95)^{1/2}(D) $\frac{1}{95}$ (E) $\frac{1}{(95)^2}$ 

$C_2H_4(g)$ is reduced by $H_2(g)$ in the presence of a solid platinum catalyst, as represented by the equation above. Factors that could affect the rate of the reaction include which of the following?

- I. Changes in the partial pressure of $H_2(g)$
- II. Changes in the particle size of the platinum catalyst
- III. Changes in the temperature of the reaction system

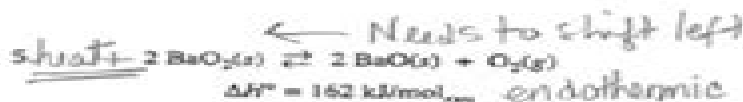
(A) III only

(B) I and II only

(C) I and III only

(D) II and III only

(E) I, II, and III



A sealed rigid vessel contains $BaO_2(s)$ in equilibrium with $BaO(s)$ and $O_2(g)$ as represented by the equation above. Which of the following changes will increase the amount of $BaO_2(s)$ in the vessel?

(A) Removing a small amount of $O_2(g)$ $R \times$ (B) Removing a small amount of $BaO(s)$ X (C) Adding He gas to the vessel X (D) Lowering the temperature L

2. For which of the following reaction mixtures at equilibrium would reducing the volume of the container at constant temperature cause the equilibrium to shift toward the products?

increase P°



shift towards fewer
+ make gas



When 4.00 mol each of $X(g)$ and $Y(g)$ are placed in a 1.00 L vessel and allowed to react at constant temperature according to the equation above, 6.00 mol of $Z(g)$ is produced. What is the value of the equilibrium constant, K_c ?

(A) 3

(B) 6

(C) 8

(D) 16

(E) 36



6. The synthesis of $CH_3OH(l)$ from $CO_2(g)$ and $H_2(g)$ is represented by the equation above. The value of K_c for the reaction at 483 K is 14.5.

increase heat shifts left, decrease K_c

Which of the following explains the effect on the equilibrium constant, K_c , when the temperature of the reaction system is increased to 650 K?

(A) K_c will increase because the activation energy of the forward reaction increases more than that of the reverse reaction.(B) K_c will increase because there are more reactant molecules than product molecules.(C) K_c will decrease because the reaction is exothermic.(D) K_c is constant and will not change.

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Peter E. Demmin



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web jun 13 2023 expandiertes polystyrol eps ist ein leichter geschlossenzelliger schaum für bauanwendungen und verpackungen der wärme isoliert und formbar ist eps wird in dämmungen und anderen produkten verwendet

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web 7 citations metrics abstract cellular polystyrene eps is the most commonly used cellular thermoplast whose main applications are insulation and packages one of its new applications is core material in building elements and this poses higher requirements for strength this is particularly evident in self supporting roof elements

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web expanded polystyrene eps is a closed cell lightweight resilient foamed plastic providing a host of unique properties expandable polystyrene is produced in a wide range of densities providing a varying range of physical properties these are matched to the various applications where the material is used to optimise its performance and strength

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web jul 27 2021 leicht stoßfest und ein hervorragender isolator eps ist ein vielseitiges verpackungsmaterial und das video erklärt wie eps der umwelt hilft

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web jan 28 2021 jugend forscht digital heute beschäftigen wir uns wieder mit styropor polystyrol wir zeigen euch wie dieses überall erhältliche material in verschiedenste formen gebracht werden kann

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web eps produkte eps steht für expandierbares polystyrol und ist besser bekannt unter dem markennamen porozell unterstützen u a energieeinsparungsmaßnahmen und ermöglichen nachhaltiges und ökologiegerechtes wachstum

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web expandierbares polystyrol eps erhält während der polymerisation zu festen kügelchen gaseinschlüsse die kügelchen werden in oktabin zum verarbeiter transportiert werden die kügelchen dort unter wasserdampf etwas über 100 c erwärmt

expandiert das gas und der thermoplastische kunststoff bläht sich auf

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web expanded polystyrene or eps is widely used commodity polymer it has been a material of choice for more than 50 years because of its versatility performance and cost effectiveness it is widely used in many everyday applications such as fish boxes bicycle helmets and insulation material

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web expandable polystyrene eps is a rigid and tough foam eps is often used for food packaging building insulation and packing material either as solid blocks formed to accommodate the item being protected or as loose fill pellets cushioning fragile items inside boxes time tested first class coperion and coperion k tron technology is

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web marktforschung zu expandierbarem polystyrol eps eps wird überwiegend von der bauindustrie zur isolation von wärme kälte oder auch schall eingesetzt ceresana hat erneut den weltmarkt für diesen leichten festen