

Worksheet #2 – Classification of Matter II Answers

Name: _____

Period: _____

Part A: Classification of Matters

1. Fill in the Blanks

A(n) element is a pure substance that is made of only one kind of atom.

The symbol for a(n) element is always one or two letters. When the symbol contains two letters, the first letter is always capitalized and the second letter is always small letter.

A(n) compound is a pure substance containing two or more elements that are chemically combined. A(n) compound is represented by a chemical formula. The elements in a(n) compound always combine in definite proportions.

A(n) mixture is made of two or more substances that are physically combined. A(n) mixture that is uniformly mixed is called homogeneous mixture.

A special name for this is a(n) solution. A(n) mixture that is not uniformly mixed is called heterogeneous mixture. A special type of mixture that is a solid solution of two or more metals is called a(n) alloy.

2. Identify each of the common items below as an element, a compound or a mixture. Their chemical formulas have been given to help you out.

- a. Sugar ($C_6H_{12}O_6$) compound
- b. Phosphorous (P_4) element
- c. Battery Acid (H_2SO_4) compound
- d. Air (N_2 & O_2 & CO_2 ...) mixture (depends)
- e. Oxygen (O_2) element
- f. Milk ($C_6H_{12}O_6$ & H_2O & $C_{50}H_{102}O_7$) heterogeneous mixture
- g. Gold (Au) element

- h. Granite (SiO_2 & $KAlSi_3O_8$ & $K_2Si_2O_7$) mixture (heterogeneous)
- i. Drain Cleaner (Al & NaOH & ...) mixture (homogeneous)
- j. Sodium (Na) element
- k. Water (H_2O) compound
- l. Salt Water (NaCl & H_2O) mixture (homogeneous)

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