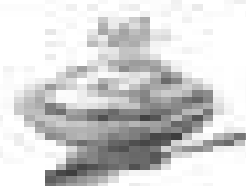


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine which elements are present in a sample of a substance by using the emission spectrum.

Part I: Finding Evidence for Discrete Energy Levels

When a substance is heated, it emits light. The light is composed of many different colors. The colors are called spectral lines. The spectral lines are the result of the emission of light by the atoms of the substance. The spectral lines are the result of the emission of light by the atoms of the substance.

Observation: The emission spectrum of hydrogen shows a series of discrete spectral lines. The spectral lines are the result of the emission of light by the atoms of the substance.

The spectral lines are the result of the emission of light by the atoms of the substance. The spectral lines are the result of the emission of light by the atoms of the substance.

Conclusion: The emission spectrum of hydrogen shows a series of discrete spectral lines. The spectral lines are the result of the emission of light by the atoms of the substance. The spectral lines are the result of the emission of light by the atoms of the substance.

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Answers Lesson 11 Atomic Pudding

Dr. Amar Bhutani



Answers Lesson 11 Atomic Pudding:

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