

Solubility Rules

Soluble compounds include ionic compounds containing ...

1. ... alkali metal ions (such as Na^+ , K^+ , etc.).
2. ... the ammonium ion (NH_4^+).
3. ... chloride, bromide, or iodide ions **EXCEPT** when paired with Ag^+ , Pb^{2+} , or Hg_2^{2+} ions.
4. ... nitrate, acetate, or chlorate ions **EXCEPT** $\text{AgC}_2\text{H}_3\text{O}_2$.
5. ... the sulfate ion **EXCEPT** when paired with Pb^{2+} , Ca^{2+} , Sr^{2+} , Ba^{2+} , or Ra^{2+} ions.

Insoluble compounds include ionic compounds containing ...

6. ... the sulfide ion **EXCEPT** when paired with alkali metal, alkaline earth metal, or ammonium ions.
7. ... the hydroxide ion **EXCEPT** when paired with alkali metal, ammonium, or Sr^{2+} , or Ba^{2+} ions.
8. ... phosphate, carbonate, or sulfite ions **EXCEPT** when paired with alkali metal or ammonium ions.

Answers Pogil Solubility Rules

U. Bendict



Answers Pogil Solubility Rules:

Answers Pogil Solubility Rules Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has are more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Answers Pogil Solubility Rules**, a literary masterpiece that delves deep in to the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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