

CHEM 151 – Quiz #4
Thermo, Redox, and Electrochem
 December 4, 2018

1. Consider this unbalanced reaction at 25 °C:

Zinc solid reacts with a solution containing nitrate ion, to form zinc ion and nitrogen monoxide gas.

The standard reduction potentials (E°) are:

Nitrogen half-reaction: $+0.957\text{ V}$ red
 Zinc half-reaction: -0.762 V ox

a. (2 pts) Using chemical formulas, write the unbalanced reaction given above.



b. (5 pts) Balance the reaction in an acidic solution using the half-reaction method practiced in class (include phases in your final balanced reaction).



c. (3 pts) i) assign oxidation numbers to all atoms in your final balanced reaction, ii) calculate the change of oxidation state for the two atoms that underwent change, and identify if each atom underwent oxidation or reduction, and iii) label the oxidizing and reducing agents (from the reactants only).

i) see above

ii) $\text{Zn} \xrightarrow{\text{iii)}} 0 \rightarrow +2$, oxidation $\Rightarrow \text{Zn(s)} = \text{reducing agent}$

$\text{N} \xrightarrow{\text{iii)}} +5 \rightarrow +2$, reduction $\Rightarrow \text{NO}_3^-(\text{aq}) = \text{oxidizing agent}$

Answers For Chemistry 151 Lab

Jessica J Manson



Answers For Chemistry 151 Lab:

This book delves into Answers For Chemistry 151 Lab. Answers For Chemistry 151 Lab is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Answers For Chemistry 151 Lab, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Answers For Chemistry 151 Lab
- Chapter 2: Essential Elements of Answers For Chemistry 151 Lab
- Chapter 3: Answers For Chemistry 151 Lab in Everyday Life
- Chapter 4: Answers For Chemistry 151 Lab in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Answers For Chemistry 151 Lab. This chapter will explore what Answers For Chemistry 151 Lab is, why Answers For Chemistry 151 Lab is vital, and how to effectively learn about Answers For Chemistry 151 Lab.

3. In chapter 2, this book will delve into the foundational concepts of Answers For Chemistry 151 Lab. The second chapter will elucidate the essential principles that must be understood to grasp Answers For Chemistry 151 Lab in its entirety.

4. In chapter 3, the author will examine the practical applications of Answers For Chemistry 151 Lab in daily life. This chapter will showcase real-world examples of how Answers For Chemistry 151 Lab can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Answers For Chemistry 151 Lab in specific contexts. This chapter will explore how Answers For Chemistry 151 Lab is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Answers For Chemistry 151 Lab. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Answers For Chemistry 151 Lab.

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