#### REVIEW OF BASIC CHEMISTRY ANSWER KEY

Name the following elements. Spelling counts:

Write the symbols for the following elements.

| H  | hydrogen |
|----|----------|
| S  | sulphur  |
| CI | chlorine |
| C  | carbon   |
| N  | nitrogen |
| Na | sodium   |
| Pb | lead     |
| Ag | silver   |
|    |          |

| sodium     | PMan |
|------------|------|
| phosphorus | P    |
| fluorine   | -    |
| magnesium  | Mg   |
| potassium  | K    |
| calcium    | Car  |
| zina       | Zn   |
| iron       | Fo   |
|            |      |

Using a periodic table, record the atomic number for the following elements:

gold (Au) 79 copper (Cu) 29

 Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (CI) 35.5 calcium (Ca) 40.1

 Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

| Element | Atomic<br>Number | Atomic<br>Mass | Protons | Neutrons | Electrons |
|---------|------------------|----------------|---------|----------|-----------|
| Lead    | 82               | 207            | 82      | 125      | 82        |
| Argon   | 18               | 58             | 18      | 40       | 18        |
| Barium  | 56               | 137            | 56      | 81       | 56        |
| Tin     | 50               | 112            | 50      | 62       | 50        |

6. Match the unit with its description:

| _F | 302 | amount of substance | A. | joule |
|----|-----|---------------------|----|-------|
| C  | 2.  | density             | в. | g     |
| В  | 3.  | mass                | G. | g/m/L |
| D  | 4.  | molar mass          | D. | g/mok |
| A_ | 5.  | energy              | €. | L     |
| E  | 6.  | volume              | F. | mole  |

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