

Practicing the Mole - - Even Problems Answer Key

Calculate the mass in grams of each of the following:

2. 8.00 moles of aluminum

$$\frac{8.00 \text{ mol} \mid 26.98 \text{ g}}{1 \text{ mol}} = 216 \text{ g Al}$$

6. 7.00 moles of iodine (I₂)

$$\frac{7.00 \text{ mol} \mid 253.8 \text{ g}}{1 \text{ mol}} = 1,780 \text{ g I}_2$$

4. 2.00×10^2 moles of chlorine (Cl₂)

$$\frac{200. \text{ mol} \mid 70.90 \text{ g}}{1 \text{ mol}} = 14,200 \text{ g Cl}_2$$

8. 9.20 moles of iron

$$\frac{9.20 \text{ mol} \mid 55.85 \text{ g}}{1 \text{ mol}} = 514 \text{ g Fe}$$

Calculate the number of moles of atoms in each of the following:

10. 281 g of iron

$$\frac{281 \text{ g} \mid 1 \text{ mol}}{55.85 \text{ g}} = 5.03 \text{ mol Fe}$$

14. 93.0 g of phosphorus

$$\frac{93.0 \text{ g} \mid 1 \text{ mol}}{30.97 \text{ g}} = 3.00 \text{ mol P}$$

12. 72.0 g of magnesium

$$\frac{72.0 \text{ g} \mid 1 \text{ mol}}{24.31 \text{ g}} = 2.96 \text{ mol Mg}$$

16. 1035 g of lead

$$\frac{1035 \text{ g} \mid 1 \text{ mol}}{207.2 \text{ g}} = 4.995 \text{ mol Pb}$$

Calculate the number of atoms in each of the following:

18. 5.40 moles of iron

$$\frac{5.40 \text{ mol} \mid 6.022 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 3.25 \times 10^{24} \text{ atoms Fe}$$

20. 10.0 moles of titanium

$$\frac{10.0 \text{ mol} \mid 6.022 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 6.02 \times 10^{24} \text{ atoms Ti}$$

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Water Chemistry Stanley E. Manahan, 2010-08-19 Carefully crafted to provide a comprehensive overview of the chemistry of water in the environment Water Chemistry Green Science and Technology of Nature's Most Renewable Resource examines water issues within the broad framework of sustainability an issue of increasing importance as the demands of Earth's human population threaten to overwhelm it General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems Chemistry Karen Timberlake, 2003 Chemistry An Introduction to General Organic and Biological Chemistry now in its eighth edition makes chemistry exciting by showing why important concepts are relevant to the lives and future careers of readers The new design digital images photos Career Focus features and macro to micro art enhance the new edition while it retains the many features that have made this book so successful The writing as always is exceptionally friendly Each section contains sample problems that develop readers critical thinking skills This edition also contains more conceptual problems than ever before and has been redesigned to accommodate new styles of learning and teaching with a wide variety of pedagogical tools Health and Environmental Notes throughout the book highlight topics that are relevant to readers lives and are ideal for classroom discussion Explore Your World activities in each chapter make chemistry exciting relevant and non threatening

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with these plans you can build these trailers in two days and then have a trailer that can be used for many years whether you re an experienced craftsman or just an amateur these improvised utility trailer plans easily adapt to your needs for example you can use them to transport items such as groceries bicycles and more diy utility trailer