
Chem - Covalent Bonding: Questions & Answers

properties of non metals ✓ Ans - inability to conduct electricity - absence of ions or delocalised electrons
low boiling and melting points - generally weak intermolecular bonds, can be soft

molecules ✓ Ans - separate groups of atoms with a formula, bonded together

intermolecular bonds ✓ Ans - bonds between molecules (dispersion, dipole-dipole, hydrogen bonding) determines physical properties

intramolecular bonds ✓ Ans - bonds between atoms in a molecule (covalent, ionic eg.)

covalent bonds ✓ Ans - non metals share electrons to complete the octet rule, an atom forms as many bonds as it needs eg. carbon needs 4 electrons so forms 4 bonds

why do non metals share electrons not transfer? ✓ Ans - non metals have a relatively high electronegativity so competing to attract each others electron, shares instead

electron dot diagrams ✓ Ans - lewis diagram shows valence electrons (bonding pairs) and lone pairs

valence structure ✓ Ans - each pair of bonding electrons is shown as a line, still represents dots as lone pairs. 1 line=2 electrons

bonding pairs and lone pairs ✓ Ans - bonding pairs; shared electrons in bond
lone pairs; non bonding pairs

double covalent bond ✓ Ans - rigid, two pairs of electrons (4 electrons) are shared, represented as a double line in valence structures eg. O_2

Answers For Chem Think Covalent Bonding

Leallyn Burr Clapp



Answers For Chem Think Covalent Bonding:

Unveiling the Power of Verbal Artistry: An Emotional Sojourn through **Answers For Chem Think Covalent Bonding**

In some sort of inundated with monitors and the cacophony of instantaneous connection, the profound energy and mental resonance of verbal beauty often diminish in to obscurity, eclipsed by the constant barrage of noise and distractions. Yet, nestled within the lyrical pages of **Answers For Chem Think Covalent Bonding**, a captivating work of fictional beauty that impulses with raw emotions, lies an unique trip waiting to be embarked upon. Penned with a virtuoso wordsmith, that exciting opus instructions visitors on an emotional odyssey, softly revealing the latent potential and profound affect embedded within the delicate web of language. Within the heart-wrenching expanse with this evocative analysis, we will embark upon an introspective exploration of the book is key themes, dissect their charming writing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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