

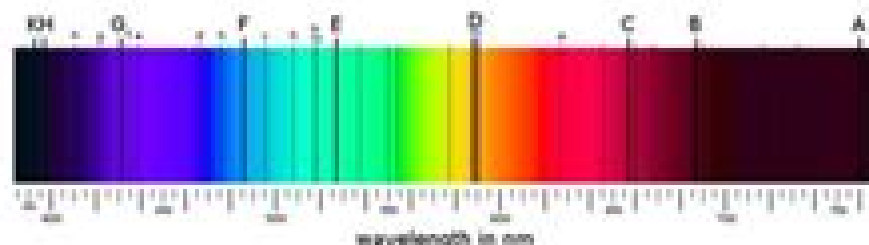
Spectral Analysis

Worksheet

Types of spectra:

Type of spectrum	Photographic example
Continuous (or continuum)	
Absorption (dark line)	
Emission (bright line)	

Here is a spectrum with many of the known spectral lines.



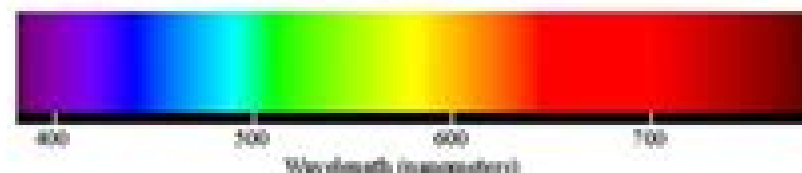
1. What colour is a photon of light with wavelength

- a. 450 nm _____ b. 700 nm _____
c. 523 nm _____ d. 560 nm _____

2. What colour is the emission of hydrogen-alpha, H α ?

3. Draw the main spectrum lines for a star containing Iron and Hydrogen.

(note: Some elements have more than one absorption line)



4. Draw the Emission Lines on a spectrum for a Magnesium lamp. Insert a wavelength values



5. A Nebula of mainly Hydrogen and Mercury (unlikely) is between a star and the earth.

a. What happens to the light as it passes through the nebula

b. The Star is mainly hydrogen and Helium. Complete the spectrum



Designation	Element	Wavelength (nm)	Designation	Element	Wavelength (nm)
y	O γ	844.663	a	Fe	485.763
Z	O γ	844.664	P	H β	486.134
A	O γ	844.665	d	Fe	486.834
B	O γ	844.666	e	Fe	488.335
C	H α	656.281	G	H γ	434.047
e	O γ	844.667	G	Fe	435.790
D γ	Na	589.592	G	Ca	435.774
D γ	Na	589.593	h	H δ	434.175
D γ or d	H α	656.282	H	Ca	394.847
e	H α	656.283	K	Ca	395.348
E γ	Fe	527.809	L	Fe	382.044
h γ	Mg	448.142	N	Fe	358.121
h γ	Mg	448.143	P	Fe	358.132
h γ	Fe	516.891	T	Fe	302.108
h γ	Fe	516.892	t	Fe	299.444
h γ	Mg	448.143			

Analysis Of Spectral Lines Answers

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Modern Spectrochemical Analysis of Metals Volker B. E. Thomsen, 1996-01-01 This book provides a fundamental introduction to spectroscopy Designed for both beginning and experienced users of spectrometers it provides the background necessary to understand the instrumentation Numerous examples and figures are utilized to illustrate the main points Most equations and formulas come with sample calculations to show actual usage Examples of authentic data from genuine spectrochemical analysis of selected metals can be found throughout Book jacket

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Journal of Research of the U.S. Geological Survey Geological Survey (U.S.), 1974

The Principles of Chemistry Dmitry Ivanovich Mendeleev, 1897

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