Gizmos: Types of Reactions

Activity B:	Get the Gizmo ready:	80 0
Classifying reactions	Turn off Show summary and Show histograms.	1

Introduction: Chemical equations show how compounds and elements react with one another. An element is a substance consisting of one kind of atom, such as aluminum (Al) or oxygen gas (O₂). A compound is a substance made of more than one kind of atom, such as water (H₂O) or table salt (NaCl).

Question: How are chemical reactions classified?

 $G. _Ag_2O \rightarrow _Ag + _O_2$

H. Al + CuCl₂ → AlCl₃ + Cu

 Match: Most chemical reactions can be classified as one of four types. Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction.

	One reactant is broken down into two or	A	Addition
	more products.		
-	Two or more reactants combine to form one product.	В.	Decomposition
	Two compounds react to form two different compounds.	C.	Single replacement
-	A compound reacts with an element to form a new compound and a different element.	D.	Double replacement

 Practice: Balance each of the chemical equations below. (Some equations may already be in balance.) In the space to the right, classify the reaction as an addition, decomposition, single replacement, or double replacement reaction.

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