



Name: _____
Date: _____

COUNTING ATOMS

WORKSHEET

- Chemical formulas can have three components.
 - The _____ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
 - The _____ which are numbers that are found on the lower _____ - hand side of each element symbol. This tells you the number of _____ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only _____ atom of this element in the molecule.
 - The _____ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be _____ by the _____ on the outside of the bracket found on the lower _____ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH	
Element	# of Atoms
Total	

4) Li ₂ SO ₄	
Element	# of Atoms
Total	

7) H ₂ COCH ₃	
Element	# of Atoms
Total	

10) NH ₄ C ₂ H ₃ O ₂	
Element	# of Atoms
Total	

13) Ca ₃ (PO ₄) ₂	
Element	# of Atoms
Total	

2) HNO ₃	
Element	# of Atoms
Total	

5) NH ₄ Cl	
Element	# of Atoms
Total	

8) Mg(OH) ₂	
Element	# of Atoms
Total	

11) NaC ₂ H ₃ O ₂	
Element	# of Atoms
Total	

14) Fe ₂ (CO ₃) ₃	
Element	# of Atoms
Total	

3) U ₂ O	
Element	# of Atoms
Total	

6) CaClO ₄	
Element	# of Atoms
Total	

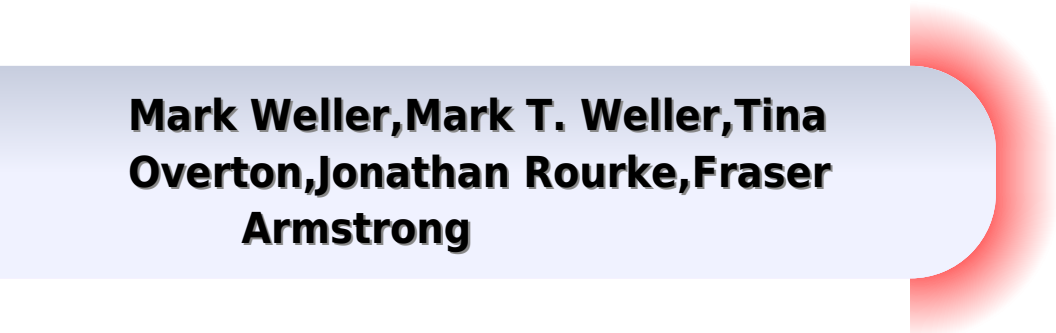
9) Al(OH) ₃	
Element	# of Atoms
Total	

12) (NH ₄) ₃ PO ₄	
Element	# of Atoms
Total	

15) Al ₂ (SO ₄) ₃	
Element	# of Atoms
Total	

Answer Key For Counting Atoms In Compound

**Mark Weller, Mark T. Weller, Tina
Overton, Jonathan Rourke, Fraser
Armstrong**



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Table of Contents Answer Key For Counting Atoms In Compound

1. Understanding the eBook Answer Key For Counting Atoms In Compound
 - The Rise of Digital Reading Answer Key For Counting Atoms In Compound
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Counting Atoms In Compound
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Counting Atoms In Compound
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Counting Atoms In Compound
 - Personalized Recommendations
 - Answer Key For Counting Atoms In Compound User Reviews and Ratings
 - Answer Key For Counting Atoms In Compound and Bestseller Lists
5. Accessing Answer Key For Counting Atoms In Compound Free and Paid eBooks

- Answer Key For Counting Atoms In Compound Public Domain eBooks
 - Answer Key For Counting Atoms In Compound eBook Subscription Services
 - Answer Key For Counting Atoms In Compound Budget-Friendly Options
6. Navigating Answer Key For Counting Atoms In Compound eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key For Counting Atoms In Compound Compatibility with Devices
 - Answer Key For Counting Atoms In Compound Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key For Counting Atoms In Compound
 - Highlighting and Note-Taking Answer Key For Counting Atoms In Compound
 - Interactive Elements Answer Key For Counting Atoms In Compound
 8. Staying Engaged with Answer Key For Counting Atoms In Compound
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Counting Atoms In Compound
 9. Balancing eBooks and Physical Books Answer Key For Counting Atoms In Compound
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Counting Atoms In Compound
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Answer Key For Counting Atoms In Compound
 - Setting Reading Goals Answer Key For Counting Atoms In Compound
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Answer Key For Counting Atoms In Compound
 - Fact-Checking eBook Content of Answer Key For Counting Atoms In Compound
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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