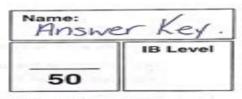
Chemical Quantities Unit Test



 Answer the following in the space provided. Where required, be sure to use correct scientific notation, significant digits and including proper units of measure.
 marks)

(12 marks)	is of measure.
(a) the product(s) of the reaction of $SO_{3(g)} + H_2O_{(l)} \rightarrow$	H2504(00)
(b) the product(s) of the reaction of $Ba_{(s)} + H_2O_{(t)} \longrightarrow$	Ba(OH)2000 + H219)
(c) the product(s) of the reaction of $Na_2CO_{3(s)}+HNO_{3(aq)} \rightarrow$	Nanogen + H2010 + CO2
(d) the product(s) of the reaction of $Ag_{(s)} + Pb(NO_3)_{2(aq)} \rightarrow$	no rxn.
(e) the number of moles in 9.97 x 10^{25} molecules of CO_2	166 mol
(f) the number of H atoms in 2.02 moles of $CH_3(CH_2)_8COOH$	2.43 × 1025 H atoms
(g) the number of moles in a 71.92 g sample of MgSO ₄	0.5974 mol
(h) the mass of 3.33 x 10 ²⁰ molecules of chlorine	0.03929
(i) percentage composition of ammonium phosphate chemical formula	(NH4) PO4
molar mass	149.12 A.I
payagataga composition	28.19% N 8.13% H
percentage composition	20.77% P 42.92% O

Answer the following with complete solutions, be sure to indicate all units of measure and express final answer to correct significant digits.

 A molecular compound containing the unknown element X has the formula X₄O₁₀. If it is known that the compound is 56.36% oxygen by mass, calculate the molar mass of element X and state it's identity. (2 marks)

$$56.36\% = \frac{10(16.00 \, \text{m}_{\odot})}{M_{\text{XuPo}}} \times 100\%$$
 $283.9 \, \text{m}_{\odot} = 4(M_{\text{X}}) + 10(16.00 \, \text{m}_{\odot})$
 $M_{\text{XuPo}} = 283.9 \, \text{m}_{\odot} = 283.9 \, \text{m}_{\odot} = 30.97 \, \text{m}_{\odot} =$

Answer Key For Chemical Quantities Test

JG Myers

Answer Key For Chemical Quantities Test:

<u>CliffsNotes AP Chemistry</u> Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

The Chemical Gazette, Or, Journal of Air Force Manual United States. Department of the Air Force,1973 Practical Chemistry, in All Its Applications to Pharmacy, Arts, and Manufactures ,1851 RCF Ltd Management Trainee Exam PDF-Rashtriya Chemicals and Fertilizers Ltd Management Trainee (Chemical) Exam-Chemical Engineering Subject Practice Sets Chandresh Agrawal, Nandini Books, 2025-01-29 SGN The RCF Ltd Management Trainee Exam PDF Rashtriya Chemicals and Fertilizers Ltd Management Trainee Chemical Exam Chemical Engineering Subject Practice Sets Covers Objective Questions With Answers Frank Schaffer's Chemistry for Everyday David Thurlo, 1999 Students learn about important subjects by relating them to events and things that occur in their everyday lives A wealth of interesting activities provide a detailed look into each subject Easy to use activities can be completed individually at school or at home though a few hands on experiments require group work and data sharing A great supplement to any existing curriculum Includes topics such as the scientific method applied to chemistry determining specific gravity balancing Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference **Inorganic Chemistry for Undergraduates** R. Gopalan, 2009 The Chemical Gazette, Or, and test preparation Journal of Practical Chemistry, in All Its Applications to Pharmacy, Arts, and Manufactures William Francis, Henry Croft, 1851 The Journal of the Society of Chemical Industry, 1892 Official Gazette of the United States Patent Office United

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