Name:	KEY	Block:

# Assignment #2 - Compound Names and Formulas Multi-valent ions only

A. Write the correct formula for the	e following compounds, all of which have been named using the
modern Roman Numeral Method.	The ionic charge is given after the first element (metallic) in Roman
Numerals.	
~ ~	

1. copper (II) oxide	11. manganese (III) oxide Mn 203
2. mercury (I) oxide	12. vanadium (II) bromide VB(z
3. gold (III) chloride _AuCl3	13. niobium(V) oxide No205
4. thallium (III) bromide TUSV 3	14. titanium (III) oxide T\20>
5. bismuth (V) oxide <u>B1205</u>	15. titanium (III) nitride TIN
6. terbium (III) oxide	16. iron (II) oxide <u>Fe.O</u>
7. uranium (VI) oxide \( \lambda \)	17. cobalt (II) phosphide C03P2
8. protactinium (IV) bromide PhBY4	18. tin (II) oxide
9. cerium (III) oxide	19. thulium (II) bromide TWBY 2
10. arsenic (V) sulphideAS2S5	20. copper (II) bromide CURY2
B. Name the compound using the Roman Numer	
1. SnCl Tin (IV) Chloride	9. PdF. Palladium (IV) Fluoride
2. BIBIS BISMINTO BIOMICE	10. Os2Os OSMIUM (III) Oxide
3. POO2 POLONIUM (IV) OXICE	11. MOBY MOLYOCHNIAM (II) Bramide
4. Pb/2 100d(II) Iodide	12. VOIS Vanadium (V) Chloride
5. Hgo Mercury (II) Oxide	13. Mn <sub>2</sub> O <sub>3</sub> Managamese (III) Oxide
6. HgCI _ MEYCUYY (I) Chroride_	
7. Augos Gold (III) Oxide	15. NP2O3 NEDTURNIUM (II) OXKIC
8. FeCla Iron (II) Chlorida	16. V2Os Vanadium (V) Oxide

# **Answer Key To Writing Ionic Formulas Math Skills**

James C. Hill

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# **Table of Contents Answer Key To Writing Ionic Formulas Math Skills**

- 1. Understanding the eBook Answer Key To Writing Ionic Formulas Math Skills
  - The Rise of Digital Reading Answer Key To Writing Ionic Formulas Math Skills
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Answer Key To Writing Ionic Formulas Math Skills
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Key To Writing Ionic Formulas Math Skills
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answer Key To Writing Ionic Formulas Math Skills
  - Personalized Recommendations
  - Answer Key To Writing Ionic Formulas Math Skills User Reviews and Ratings
  - Answer Key To Writing Ionic Formulas Math Skills and Bestseller Lists
- 5. Accessing Answer Key To Writing Ionic Formulas Math Skills Free and Paid eBooks
  - Answer Key To Writing Ionic Formulas Math Skills Public Domain eBooks
  - Answer Key To Writing Ionic Formulas Math Skills eBook Subscription Services
  - Answer Key To Writing Ionic Formulas Math Skills Budget-Friendly Options

- 6. Navigating Answer Key To Writing Ionic Formulas Math Skills eBook Formats
  - o ePub, PDF, MOBI, and More
  - Answer Key To Writing Ionic Formulas Math Skills Compatibility with Devices
  - Answer Key To Writing Ionic Formulas Math Skills Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answer Key To Writing Ionic Formulas Math Skills
  - Highlighting and Note-Taking Answer Key To Writing Ionic Formulas Math Skills
  - Interactive Elements Answer Key To Writing Ionic Formulas Math Skills
- 8. Staying Engaged with Answer Key To Writing Ionic Formulas Math Skills
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answer Key To Writing Ionic Formulas Math Skills
- 9. Balancing eBooks and Physical Books Answer Key To Writing Ionic Formulas Math Skills
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Key To Writing Ionic Formulas Math Skills
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key To Writing Ionic Formulas Math Skills
  - Setting Reading Goals Answer Key To Writing Ionic Formulas Math Skills
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key To Writing Ionic Formulas Math Skills
  - Fact-Checking eBook Content of Answer Key To Writing Ionic Formulas Math Skills
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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