

Grade 11 Chemistry

INTERMOLECULAR FORCES & Physical Properties

EXAM QUESTIONS PART 3



NAME OF SUBSTANCE	FORMULA	BOILING POINT (°C)
Water	H ₂ O	100
Ethanol	CH ₃ CH ₂ OH	78
Bromine	Br ₂	58,8
Iodine	I ₂	184,3
Ammonia	NH ₃	-33,3
Phosphine	PH ₃	-87,7

Explain why ethanol is soluble in water. Refer to the relative strength of the intermolecular forces in ethanol and water.

Explain why the boiling point of iodine is higher than that of bromine. Refer to the intermolecular forces present in EACH substance in the explanation.

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Table of Contents Answers For Grade 11 Intermolecular Formal Task

1. Understanding the eBook Answers For Grade 11 Intermolecular Formal Task
 - The Rise of Digital Reading Answers For Grade 11 Intermolecular Formal Task
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Grade 11 Intermolecular Formal Task
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Grade 11 Intermolecular Formal Task
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Grade 11 Intermolecular Formal Task
 - Personalized Recommendations
 - Answers For Grade 11 Intermolecular Formal Task User Reviews and Ratings
 - Answers For Grade 11 Intermolecular Formal Task and Bestseller Lists

5. Accessing Answers For Grade 11 Intermolecular Formal Task Free and Paid eBooks
 - Answers For Grade 11 Intermolecular Formal Task Public Domain eBooks
 - Answers For Grade 11 Intermolecular Formal Task eBook Subscription Services
 - Answers For Grade 11 Intermolecular Formal Task Budget-Friendly Options
6. Navigating Answers For Grade 11 Intermolecular Formal Task eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Grade 11 Intermolecular Formal Task Compatibility with Devices
 - Answers For Grade 11 Intermolecular Formal Task Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Grade 11 Intermolecular Formal Task
 - Highlighting and Note-Taking Answers For Grade 11 Intermolecular Formal Task
 - Interactive Elements Answers For Grade 11 Intermolecular Formal Task
8. Staying Engaged with Answers For Grade 11 Intermolecular Formal Task
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Grade 11 Intermolecular Formal Task
9. Balancing eBooks and Physical Books Answers For Grade 11 Intermolecular Formal Task
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Grade 11 Intermolecular Formal Task
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For Grade 11 Intermolecular Formal Task
 - Setting Reading Goals Answers For Grade 11 Intermolecular Formal Task
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Grade 11 Intermolecular Formal Task
 - Fact-Checking eBook Content of Answers For Grade 11 Intermolecular Formal Task
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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