



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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1. Understanding the eBook Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - The Rise of Digital Reading Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Personalized Recommendations

- Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 User Reviews and Ratings
- Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 and Bestseller Lists
- 5. Accessing Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 Free and Paid eBooks
 - Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 Public Domain eBooks
 - Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 eBook Subscription Services
 - Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 Budget-Friendly Options
- 6. Navigating Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 Compatibility with Devices
 - Answers Of Grade 11 Physical Sciences Test 1 20 03 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Highlighting and Note-Taking Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Interactive Elements Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
- 8. Staying Engaged with Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
- 9. Balancing eBooks and Physical Books Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Setting Reading Goals Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers Of Grade 11 Physical Sciences Test 1 20 03 2014
 - Fact-Checking eBook Content of Answers Of Grade 11 Physical Sciences Test 1 20 03 2014

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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