

Chapter 1 – Introduction to Chemistry – Answer Key

Chapter 1: 1, 3, 5, 8, 9, 10, 12 - 19, 21, 22, 30, 36 (18 total)

Section Review 1.1

1. Match the numbered terms in the right column with the lettered terms in the left column.

__4__	a.	Technology	1.	Life
__3__	b.	Organic chemistry	2.	Matter and its changes
__1__	c.	Biochemistry	3.	Carbon
__2__	d.	Chemistry	4.	Applications
__5__	e.	Analytical chemistry	5.	Composition

3. Explain the difference between *pure* and *applied* chemistry.

Pure chemistry examines chemical aspects of the world for the sake of knowledge alone; applied chemistry uses this knowledge to solve real-world problems.

Section Review 1.2

5. Name at least three areas of science in which chemistry plays an important role.

- a. agriculture
- b. medical and biotechnological sciences
- c. materials science

Section Review 1.3

8. Describe the steps involved in the scientific method. Provide an example.

Observation: My stomach is growling

Hypothesis: Maybe I'm hungry

Experiment: Eat something and see what happens

9. Distinguish between a *theory*, a *hypothesis*, and a *scientific law*.

A law is a concise statement that summarizes the results of a broad variety of observations and experiments. A hypothesis is a proposed explanation or reason for what is observed. A theory is an extensively tested explanation for experimental results.

10. Why should a hypothesis be developed *before* experiments take place?

The design of the experiments will be guided by the proposed hypothesis

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J Elliott



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