

**20.1****DESCRIBING ACIDS AND BASES****SECTION REVIEW****Objectives**

- List the properties of acids and bases.
- Name an acid or base when given the formula.

**Key Terms**

- acid
- base

**Part A Completion**

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

Aqueous solutions of \_\_\_\_\_ 1 \_\_\_\_\_ are electrolytes. Acids react with many \_\_\_\_\_ 2 \_\_\_\_\_ to produce \_\_\_\_\_ 3 \_\_\_\_\_ gas. \_\_\_\_\_ 4 \_\_\_\_\_ are compounds that react with acids to form \_\_\_\_\_ 5 \_\_\_\_\_ and a salt.

Acids produce \_\_\_\_\_ 6 \_\_\_\_\_ ions when dissolved in water. There is a standard system for naming acids. For example,  $\text{HBr}$  is called \_\_\_\_\_ 7 \_\_\_\_\_ acid, while  $\text{HNO}_3$  is called \_\_\_\_\_ 8 \_\_\_\_\_ acid.

Bases produce \_\_\_\_\_ 9 \_\_\_\_\_ ions when dissolved in water. Basic compounds that are \_\_\_\_\_ 10 \_\_\_\_\_ are named in the same way as other ionic compounds.

1. \_\_\_\_\_

2. \_\_\_\_\_

3. \_\_\_\_\_

4. \_\_\_\_\_

5. \_\_\_\_\_

6. \_\_\_\_\_

7. \_\_\_\_\_

8. \_\_\_\_\_

9. \_\_\_\_\_

10. \_\_\_\_\_

**Part B True-False**

Circle the word of these classes as true or false: true, AC; not true or false, NCF; or never true, NT.

- \_\_\_\_\_ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- \_\_\_\_\_ 12. An acid contains one or more hydroxide ions.
- \_\_\_\_\_ 13. Chemists have a special system for naming bases.

# 20 1 Describing Acids And Bases Answers

**Trace Jordan, Neville R. Kallenbach**



## 20 1 Describing Acids And Bases Answers:

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I. A Study of the Acid-base Equilibria of Arsphenamine Solutions Elias Elvove, 1924

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