

20.1**DESCRIBING ACIDS AND BASES****SECTION REVIEW****Objectives**

- List the properties of acids and bases.
- Name an acid or base when given the formula.

Key Terms

- acid
- base

Part A Completion

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

Aqueous solutions of _____ 1 _____ are electrolytes. Acids react with many _____ 2 _____ to produce _____ 3 _____ gas. _____ 4 _____ are compounds that react with acids to form _____ 5 _____ and a salt.

Acids produce _____ 6 _____ ions when dissolved in water. There is a standard system for naming acids. For example, HBr is called _____ 7 _____ acid, while HNO_3 is called _____ 8 _____ acid.

Bases produce _____ 9 _____ ions when dissolved in water. Some compounds that are _____ 10 _____ are named in the same way as other ionic compounds.

1. _____

2. _____

3. _____

4. _____

5. _____

6. _____

7. _____

8. _____

9. _____

10. _____

Part B True-False

Circle the word of these classes as true or false: true, AC; not true or false, NCF; or never true, NT.

- _____ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- _____ 12. An acid contains one or more hydroxide ions.
- _____ 13. Chemists have a special system for naming bases.

20 Describing Acids And Bases Answers

David A. Ucko



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Basics for Chemistry David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

Physical Chemistry and Its Biological Applications Wallace Brey, 2012-12-02 Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes Chapters 6 to 8 look at acid base equilibria and the link between electromagnetic radiation and the structure of atoms Chapters 9 to 11 cover different types of bonding the rates of chemical reactions and the process of adsorption Chapters 12 to 14 present molecular aggregates magnetic resonance spectroscopy and photochemistry and radiation This book is useful to biological scientists for self study and reference With modest additions of mathematical material by the teacher the book should also be suitable for a full year major s course in physical chemistry

Excel HSC Chemistry Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every single dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured This guide comes in a brand new format that makes even better use of your study time up to date coverage of the core topics plus 3 Option topics Industrial Chemistry Shipwrecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to section the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second hand and Investigations and Problem Solving %this way you will be able to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for word %this way you can easily match the Excel guide to the syllabus an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work hundreds of key concept questions with answers %questions that test your recall of knowledge in each chapter HSC type questions for every section in each chapter with clock icons to tell you how much time you will have to answer the questions in the HSC %this way you can test

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