

Analytical Chemistry MCQs: TYBSC SEM III

8. The solubility product K_{sp} of sparingly soluble Ag_2CrO_4 is 4×10^{-12} . The molar solubility of the salt is
- (a) 1.0×10^{-4} moles/lit
 - (b) 2.0×10^{-4} moles/lit
 - (c) 1.0×10^{-5} moles/lit
 - (d) 1.2×10^{-12} moles/lit
9. Solubility product of AgCl is 1.44×10^{-4} moles/lit at 373K. The solubility of the AgCl in boiling of water will be....
- (a) 0.72×10^{-4} moles/lit
 - (b) 0.72×10^{-2} moles/lit
 - (c) 1.20×10^{-2} moles/lit
 - (d) 1.20×10^{-4} moles/lit
10. Precipitation takes place when the ionic product....
- (a) Is zero
 - (b) Exceeds the solubility product
 - (c) Equal to solubility product
 - (d) Is less than the solubility product
11. The electromotive force is expressed in the unit of
- (a) Ohm
 - (b) Volt
 - (c) Ampere
 - (d) Coulomb
12. If the cell can provide electrical energy to an external system, it is called as
- (a) Galvanic cell
 - (b) Electrolytic cell
 - (c) Concentration cell
 - (d) Chemical cell
13. The current per unit area of electrode surface is known as....
- (a) Current efficiency
 - (b) Current yield
 - (c) Current density
 - (d) None of these

Analytical Chemistry Questions And Answers

Mitchell D. Erickson



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Quality Control in Analytical Chemistry G. Kateman, L. Buydens, 1993-09-06 Describes the basics of analytical techniques sampling and data handling in order to improve quality control in analytical laboratory management Stresses what quality parameters can be improved and which ones should be rectified first This edition includes numerous modern methods and the latest developments in time proven techniques *Basic Analytical Chemistry* Mr. Rohit Manglik, 2024-05-18 EduGorilla

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is the only chemical analysis text with an emphasis on active learning giving students step by step guidance on how the key principles of analytical science are applied in a range of practical real world contexts *Analytical Chemistry for Technicians, Second Edition* John Kenkel, 1994-07-22 The second edition of Analytical Chemistry for Technicians provides the nuts and bolts of analytical chemistry and focuses on the practical aspects for training a technician level laboratory worker This edition presents new and expanded chapters innumerable questions and problems and modified experiments that present a fresh and challenging approach Some of the topics that have been expanded include chemical equilibrium chromatography Kjeldahl method and molarity and moles where EDTA and water hardness calculations are concerned New discussions of the Ag AgCl and combination pH electrodes have been added while the discussion of ion selective electrodes has been expanded The chapter introducing instrumental analysis and computers now includes discussions of y mx b and the method of least squares The book also includes discussions of FTIR topics of NMR and mass spectrometry which are found in the new infrared spectrometry chapter **Handbook of Food Analytical Chemistry, Volume 1** Ronald E. Wrolstad, Terry E. Acree, Eric A. Decker, Michael H. Penner, David S. Reid, Steven J. Schwartz, Charles F. Shoemaker, Denise M. Smith, Peter Sporns, 2005-09-16 Emphasizing effective state of the art methodology and written by recognized experts in the field the Handbook of Food Analytical Chemistry is an indispensable reference for food scientists and technologists to enable successful analysis Provides detailed reports on experimental procedures Includes sections on background theory and

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