Class: IX

Subject: N. Science(086)

Duration: 3 hours

Maximum Marks: 80

General Instructions:

Read the following instructions very carefully and strictly follow them:

- (i) This question paper comprises 39 questions. All questions are compulsory.
- (ii) This question paper is divided into five sections A, B, C, D and E.
- (iii) Section A Questions No. 1 to 20 are multiple choice questions. Each question carries I mark.
- (iv) Section B Questions No. 21 to 26 are very short answer type questions. Each question carries 2 marks. Answer to these questions should be in the range of 30 to 50 words.
- (v) Section C Questions No. 27 to 33 are short answer type questions. Each question carries 3 marks. Answer to these questions should in the range of 50 to 80 words.
- (vi) Section D Questions No. 34 to 36 are long answer type questions. Each question carries 5 marks. Answer to these questions should be in the range of \$0 to 120 words.
- (vii) Section E Questions No. 37 to 39 are of 3 source-based/case-based units of assessment carrying 4 marks each with sub-parts.
- (viii) There is no overall choice. However, an internal choice has been provided in some sections. Only one of the alternatives has to be attempted in such questions.

			SECTION-A		
Q. No.	Questions The similarity between solid, liquid and gas matter is that, they: a) occupy space b) have mass c) both a and b d) have fixed shape				Marks 1
1					
2	When 20 g of potassium permanganate is dissolved in 1000ml of water, there is no increase in volume. This observation indicates that: a) particles of water have spaces between them into which potassium permanganate particles fit. b) particles of water attract each other. c) particle of potassium permanganate attracts each other. d) particles are continuously moving.				1
3	Which of the follow a) mist	ing is not a Liquid-ga b) clouds	s Aerosol: c) smoke	d) fog	1
-4	The ions present in Na ₂ CO ₃ are:				1

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Table of Contents 14 November Science Question Paper For Grade 9

- 1. Understanding the eBook 14 November Science Question Paper For Grade 9
 - The Rise of Digital Reading 14 November Science Question Paper For Grade 9
 - Advantages of eBooks Over Traditional Books
- 2. Identifying 14 November Science Question Paper For Grade 9
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - o Features to Look for in an 14 November Science Question Paper For Grade 9
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from 14 November Science Question Paper For Grade 9
 - Personalized Recommendations
 - 14 November Science Question Paper For Grade 9 User Reviews and Ratings
 - 14 November Science Question Paper For Grade 9 and Bestseller Lists
- 5. Accessing 14 November Science Question Paper For Grade 9 Free and Paid eBooks
 - 14 November Science Question Paper For Grade 9 Public Domain eBooks
 - 14 November Science Question Paper For Grade 9 eBook Subscription Services
 - 14 November Science Question Paper For Grade 9 Budget-Friendly Options
- 6. Navigating 14 November Science Question Paper For Grade 9 eBook Formats

- o ePub, PDF, MOBI, and More
- 14 November Science Question Paper For Grade 9 Compatibility with Devices
- 14 November Science Question Paper For Grade 9 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 14 November Science Question Paper For Grade 9
 - Highlighting and Note-Taking 14 November Science Question Paper For Grade 9
 - Interactive Elements 14 November Science Question Paper For Grade 9
- 8. Staying Engaged with 14 November Science Question Paper For Grade 9
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 14 November Science Question Paper For Grade 9
- 9. Balancing eBooks and Physical Books 14 November Science Question Paper For Grade 9
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection 14 November Science Question Paper For Grade 9
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 14 November Science Question Paper For Grade 9
 - Setting Reading Goals 14 November Science Question Paper For Grade 9
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 14 November Science Question Paper For Grade 9
 - Fact-Checking eBook Content of 14 November Science Question Paper For Grade 9
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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