

11.5

REACTIONS IN AQUEOUS SOLUTION

Section Review

Objectives

- Identify the substances that are soluble and insoluble.
- Write the balanced chemical equations for precipitation reactions.

Vocabulary

- soluble substance
- insoluble salt
- precipitate

Part A Completion

Write the complete balanced chemical equations for reactions of the reactants and products in parentheses. Assume that reactions occur whenever a precipitate is formed. Write the complete ionic equation.

- Write complete balanced equations for (a) _____, (b) _____.
- Substance A is precipitated by Substance B. Substance C is added to the mixture. _____, _____ solution.
- Write complete balanced equations for reactions of (a) _____, (b) _____.
- Write complete balanced equations for reactions of (a) _____, (b) _____.
- Write complete balanced equations for reactions of (a) _____, (b) _____.
- Write complete balanced equations for reactions of (a) _____, (b) _____.
- Write complete balanced equations for reactions of (a) _____, (b) _____.
- Write complete balanced equations for reactions of (a) _____, (b) _____.
- Write complete balanced equations for reactions of (a) _____, (b) _____.
- Write complete balanced equations for reactions of (a) _____, (b) _____.

Write a balanced chemical equation for the reaction of (a) _____, (b) _____.

Write a balanced chemical equation for the reaction of (a) _____, (b) _____.

11.1	soluble
11.2	insoluble
11.3	precipitate
11.4	precipitation reaction
11.5	precipitation reaction
11.6	precipitation reaction
11.7	precipitation reaction
11.8	precipitation reaction
11.9	precipitation reaction
11.10	precipitation reaction

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