

Buffers Keep the Balance

Post-Lab Questions

1. Compare the measured pH value for the model carbonate blood buffer to
 - a. the expected pH of an ideal carbonic acid-bicarbonate buffer, and
 - b. the actual pH of the carbonic acid-bicarbonate buffer system present in the blood.
2. Based on the pH comparisons in Question #1, which solution, the model carbonate blood buffer or an actual blood buffer, is more likely to contain a greater proportion of the carbonic acid component of the buffer compared to the bicarbonate component? Explain your reasoning.
3. What are the effects of adding even a small amount of HCl or NaOH on the pH value of the control solution (water)? Compare this to the effect of adding HCl or NaOH to the model carbonate blood buffer.
4. Which phosphate buffer solution in Part B corresponds to the composition of an ideal buffer solution? Compare its measured pH value with the calculated pH of the ideal buffer solution.

Answers Buffers Keep The Balance Lab Answers

JL Elias



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Biochemistry in the Lab Benjamin F. Lasseter, 2019-09-30 Most lab manuals assume a high level of knowledge among biochemistry students as well as a large amount of experience combining knowledge from separate scientific disciplines Biochemistry in the Lab A Manual for Undergraduates expects little more than basic chemistry It explains procedures clearly as well as giving a clear explanation of the theoretical reason for those steps Key Features Presents a comprehensive approach to modern biochemistry laboratory teaching together with a complete experimental experience Includes chemical biology as its foundation teaching readers experimental methods specific to the field Provides instructor experiments that are easy to prepare and execute at comparatively low cost Supersedes existing older texts with information that is adjusted to modern experimental biochemistry Is written by an expert in the field This textbook presents a foundational approach to modern biochemistry laboratory teaching together with a complete experimental experience from protein purification and characterization to advanced analytical techniques It has modules to help instructors present the techniques used in a time critical manner as well as several modules to study protein chemistry including gel techniques enzymology crystal growth unfolding studies and fluorescence It proceeds from the simplest and most important techniques to the most difficult and specialized ones It offers instructors experiments that are easy to prepare and execute at comparatively low cost

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