

Unit 1 - Hi! (pp. 4-7)

- 1** 1 artist 5 astronaut
2 farmer 6 teacher
3 doctor 7 musician
4 pilot 8 postman
- 2** 1 old 5 name
2 years old 6 favourite
3 job 7 capital
4 student
- 3** 2 f 4 c 6 h 8 g
3 a 5 b 7 e
- 4** a four f forty-two
b sixteen g sixty
c twenty-eight h seventy-seven
d nine i eleven
e thirty-three j fifty-five
- 5** 2 American 7 Egyptian
3 French 8 Bulgarian
4 Swiss 9 Russian
5 Spanish 10 Chinese
6 Polish
- 6** 3 ...forty years old.
4 ...eighty years old.
5 He's fifty years old.
6 He's twenty-six years old.
7 She's twelve years old.
8 He's thirty-five years old.
9 She's fifteen years old.
10 She's eighteen years old.
- 7** 1 Greece 6 the Czech Republic
2 Italy 7 Spain
3 Switzerland 8 Germany
4 Portugal 9 Hungary
5 Poland
- 8** 1 an 4 a 7 a 10 a
2 an 5 a 8 a 11 a
3 a 6 an 9 a 12 an

9	FULL FORM	SHORT FORM
	I am	I'm
	you are	you're
	he is	he's
	she is	she's
	it is	it's
	we are	we're
	you are	you're
	they are	they're

- 10** 2 am 4 are 6 am 8 are
3 is 5 are 7 is
- 11** 1 I 3 She 5 they
2 they, They 4 He 6 it
- 12** 1 Is, isn't, is/s 4 Is, isn't, is/s
2 is/s, is 5 are, am/m
3 Are, aren't, are/re
- 13** 1 What is/s your name?
2 How old are you?
3 Where are you from?
4 What is/s your job?
5 How old are you?
- 14** C: Pleased to meet you, too.
B: Pleased to meet you.
- 15** 1 I am thirty years old.
2 His favourite sport is golf.
3 The Pyramids are in Egypt.
4 Tony is sixteen years old.
5 She is from Budapest.
- 16** 1 F 3 F 5 F
2 T 4 T 6 F
- 17** (Suggested answer)

This is Juliet. She's from Paris, France. Juliet is twenty-two years old. She's a waitress. Her favourite sport is tennis.

Unit 2 - A friend in Need ... (pp. 8-11)

- 1** 1 yellow 2 white 3 green 4 black,
- 2** a) 1 head 5 toe
2 chest 6 arm
3 hand 7 leg
4 finger 8 foot
- b) (Ss' own answers)
- 3** a) Ss colour people's eyes and hair accordingly.
b) SA: What colour eyes has Ann got?
SB: She's got blue eyes.
SA: What sort of hair has she got?
SA: She's got short grey curly hair.
SA: What colour eyes has George got?
SB: He's got brown eyes.
SA: What sort of hair has he got?
SB: He's got short black hair.

Answers For French 2 Workbook

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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^- + \text{Au} \rightarrow 2\text{e}^- + 2\text{H}^+ + \text{O}_2 + \text{H}_2\text{O}_2$ 2 make electrons equal convert to basic solution $4\text{CN}^- + 2\text{Au} + 2\text{OH}^- \rightarrow 2\text{e}^- + 2\text{H}_2\text{O} + \text{O}_2 + \text{H}_2\text{O}_2 + 2\text{OH}^-$ 3 the final answer

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web aug 29 2023 the first step to balance any redox reaction is to separate the reaction into half reactions the substance being reduced will have electrons as reactants and the oxidized substance will have electrons as products usually all reactions are written as reduction reactions in half reaction tables

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bromate v ions

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation ni s ni a 2 a q 2 e now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balance each half reaction in basic solution 4 5 6 7 $\text{Cr}_2\text{O}_7^{2-} + 2\text{NO}_3^- + 2\text{MnO}_2 + \text{Cr}_3 + \text{NO}_3 + \text{SO}_2 + \text{Mn}_2\text{O}_3$ balance each redox reaction in acid solution using the half reaction method 8 $\text{H}_2\text{O}_2 + \text{Cr}_2\text{O}_7^{2-} + 2\text{H}^+ \rightarrow \text{TeO}_3 + 2\text{N}_2\text{O}_4 + \text{O}_2 + \text{Cr}_3 + \text{Te} + 10\text{H}^+ + 11\text{PbO}_2 + \text{IO}_2 + \text{I}_2 + \text{Pb}_2 + 12\text{H}^+$ as $\text{NO}_3 + \text{IO}_3 + \text{Re} + \text{IO}_3 + \text{H}_2\text{AsO}_4$

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{HNO}_2 \rightarrow \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 no 2 and $\text{C}_6\text{H}_8\text{O}_6$ b NO_2 and $\text{C}_6\text{H}_8\text{O}_6$ h and NO_2 c h and