

Chemistry of Life

DOODLE NOTES

chemistry of life
BASIC CHEMISTRY REVIEW

atoms and life
All living things are made of atoms.
An atom is the smallest part of an element. Many different types of atoms exist and they are arranged in the Periodic table of elements. Atoms are made up of subatomic particles.

Label the structure of this oxygen atom:

Electron shell
Electron
Nucleon
Proton
Neutron

• State the charge and mass of each subatomic particle.

	Proton	Neutron	Electron
Charge	+	0	-
Mass	1	1	1/1836

• How many electrons are in the first shell? 2
• How many electrons are in the outer shell? 6
• What do we call electrons in the outer shell? Valence electrons

elements
An element is a substance that is made of one type of atom only. They cannot be broken down into simpler substances. There are 118 elements. There are 92 in the periodic table.

compounds
A compound is made of two or more types of atoms that are chemically bonded. They can be broken down chemically. There are 100 compounds. There are 100 in the periodic table.

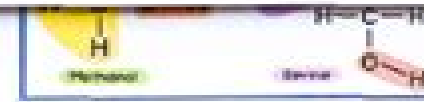
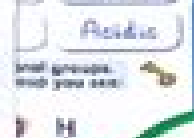
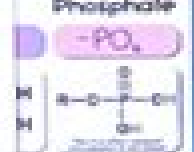
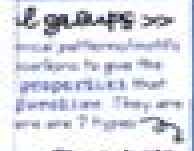
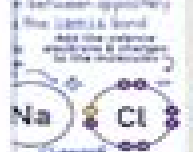
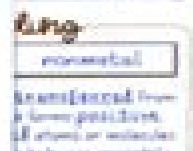
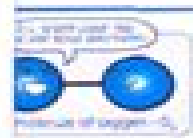
mixtures
A mixture is a mix of two or more types of atoms that are not chemically bonded. They can be separated physically. There are 100 mixtures. There are 100 in the periodic table.

Hydrogen
Hydrogen is the simplest element. It is made of one type of atom only. It is a gas at room temperature. It is used in many chemical reactions.

Helium
Helium is a noble gas. It is made of one type of atom only. It is a gas at room temperature. It is used in many chemical reactions.

Sodium Chloride
Sodium Chloride is a compound. It is made of two types of atoms: Sodium and Chlorine. It is a solid at room temperature. It is used in many chemical reactions.

Oxide
Oxide is a compound. It is made of two types of atoms: Oxygen and another element. It is a solid at room temperature. It is used in many chemical reactions.



MULTIPLE VERSIONS & EXAMPLE

Answers Review Basic Units Of Life

Marcel A. Müller



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