Experiment 3

Advance Study Assignment: Resolution of Matter into Pure Substances, II. Fractional Crystallization

- 1. Using Figure 3.1, determine
 - a. the number of grams of CuSO₄ · 5H₂O that will dissolve in 100 g of H₂O at 100°C.

(6355) + (3206) + (4×10) = 159.60 o 6504

180.0 E Cuso, - 5H,0

the number of grams of water required to dissolve 4.0 g of CuSO₄ · 5H₂O at 100°C. (Hint: your answer to Problem 1(a) gives you the needed conversion factor for g CuSO₄ · 5H₂O to g H₂O₄)

c. the number of grams of water required to dissolve 18 g KNO3 at 100°C.

18gKN03 x 100.0gHz0 = 72.0 5Hz0

72.0 2000 gH₂0

the number of grams of water required to dissolve a mixture containing 18 g KNO₃ and 4.0 g CuSO₄
 5 H₂O₂ assuming that the solubility of one substance is not affected by the presence of another.

18gKNO3x -

____ g H₂O

Advance Study Assignment 30 Answers

LL Leslie

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