

Chapter 16 Thermal Energy and Heat

Section 16.1 Thermal Energy and Matter

(pages 474–478)

This section defines heat and describes how work, temperature, and thermal energy are related to heat. It also discusses thermal expansion and contraction of materials, and explains uses of a calorimeter.

Reading Strategy (page 474)

Previewing Before you read, preview the figures in this section and add two more questions to the table. As you read, write answers to your questions. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Thermal Energy and Matter	
Questions About Thermal Energy and Matter	Answers
Which has more thermal energy, a cup of tea or a pitcher of juice?	A pitcher of juice
Why did Rumford conclude that heat is not a form of matter? (Fig. 1)	The brass was hot enough to make water boil only during drilling, so the heat must be related to the motion of the drill.
How is specific heat related to temperature? (Fig. 3)	The lower a material's specific heat, the more its temperature rises when a given amount of energy is absorbed by a given mass.

Work and Heat (page 474)

- Heat is the transfer of thermal energy from one object to another as the result of a difference in _____. Circle the correct answer.
density potential energy temperature
- Circle the letter of each sentence that is true about heat.
 - Heat is a fluid that flows between particles of matter.
 - b Heat flows spontaneously from hot objects to cold objects.
 - c Friction produces heat.

Temperature (page 475)

- What is temperature? Temperature is a measure of how hot or cold an object is compared to a reference point.

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Michael J. Padilla



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1. Understanding the eBook Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - The Rise of Digital Reading Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Personalized Recommendations

- Answer Key To Temperature Thermal Energy And Heat Pearson Education User Reviews and Ratings
- Answer Key To Temperature Thermal Energy And Heat Pearson Education and Bestseller Lists
- 5. Accessing Answer Key To Temperature Thermal Energy And Heat Pearson Education Free and Paid eBooks
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 - Answer Key To Temperature Thermal Energy And Heat Pearson Education eBook Subscription Services
 - Answer Key To Temperature Thermal Energy And Heat Pearson Education Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Answer Key To Temperature Thermal Energy And Heat Pearson Education Compatibility with Devices
 - Answer Key To Temperature Thermal Energy And Heat Pearson Education Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Highlighting and Note-Taking Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Interactive Elements Answer Key To Temperature Thermal Energy And Heat Pearson Education
- 8. Staying Engaged with Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Temperature Thermal Energy And Heat Pearson Education
- 9. Balancing eBooks and Physical Books Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Temperature Thermal Energy And Heat Pearson Education
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Setting Reading Goals Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key To Temperature Thermal Energy And Heat Pearson Education
 - Fact-Checking eBook Content of Answer Key To Temperature Thermal Energy And Heat Pearson Education

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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