

Are All Equilibria Created Equal?

Two bottles of soda are sitting in the refrigerator. One is almost full. The other is almost empty. Slowly, the sodas are going flat. This is because the dissolved carbon dioxide is coming out of solution and going into the air space above. Of course, if you keep the bottle closed tightly, the soda doesn't go flat as fast. If the gas can't escape, it soon reaches equilibrium with the solution below. But are all equilibria the same? Does the almost empty bottle retain as much carbonation as the almost full bottle? The soda with the larger air space has more room for the carbon dioxide gas to spread out. As a result the gas particles are under lower pressure and have a lower concentration. This makes it easier for them to come out of solution. When the almost empty bottle reaches equilibrium, it will have more carbon dioxide in the space above the liquid and less carbon dioxide dissolved than the almost full bottle has. It will be flat.

Equilibrium doesn't mean equal amounts of reactants and products. It means the reactants are turning into products at the same rate that the products are turning back to reactants. Many factors influence equilibrium. These include temperature, pressure, and concentration.



When sodas discuss solution equilibrium

Answer the questions below based on the reading above, and on your knowledge of chemistry.

1. A reversible reaction is at equilibrium. The forward reaction is exothermic.
 - a. Is the reverse reaction exothermic or endothermic? _____
 - b. Which has the higher activation energy, the forward reaction or the reverse? _____
 - c. What effect does raising the temperature have on the speed of reaction for the forward reaction? Why? _____

 - d. What effect does raising the temperature have on the speed of reaction for the reverse reaction? Why? _____

 - e. Will the rate of the forward reaction and the reverse reaction be affected to the same extent by an increase in temperature? Explain. _____

Are All Equilibria Created Equal Answers

E. Atlee Jackson



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CIRCULAR FLOW OF INCOME 4 CONCEPTS AND AGGREGATES RELATED TO NATIONAL INCOME 5 MEASUREMENT OF NATIONAL INCOME 6 MONEY MEANING EVOLUTION AND FUNCTION 7 COMMERCIAL BANKS AND CREDIT CREATION 8 CENTRAL BANK MEANING AND FUNCTIONS 9 INDIAN BANKING SYSTEM RECENT REFORMS AND ISSUES 10 AGGREGATE DEMAND AND AGGREGATE SUPPLY 11 DETERMINATION OF EQUILIBRIUM LEVEL OF INCOME EMPLOYMENT AND OUTPUT 12 INVESTMENT MULTIPLIER AND ITS WORKING 13 PROBLEMS OF DEFICIENT AND EXCESS DEMAND 14 MEASURES TO CORRECT DEFICIENT EXCESS DEMAND 15 GOVERNMENT BUDGET AND THE ECONOMY 16 FOREIGN EXCHANGE RATE 17 BALANCE OF PAYMENTS Economics Model Paper Chapter wise Question Answer With Marking Scheme Class XII SBPD Editorial Board,2020-11-07 1 100% Based on NCERT Guidelines 2 Important questions have been include chapterwise and unitwise 3 Previous year questions with answers of board examinations have been included 4 Solved Model Test Papers for board examination preparation for the current year have been included Part A Introductory Micro Economics 1 Introduction 2 Consumer Behavior and Demand 3 Producer Behavior and Supply 4 From of Market and Price Determination 5 Simple Application of Tools of Demand and Supply Curves Part B Introductory Macro Economics 6 Concepts and Aggregates Related to National Income 2 Money and Banking 3 Determination of Income and Employment 4 Government Budget and the Economy 5 Balance of Payment and Exchange Rate Model Paper Set I IV Board Examination Paper **Equilibrium Statistical Mechanics** E. Atlee Jackson,2012-11-21 Key features include an elementary introduction to probability distribution functions and uncertainty a review of the concept and significance of energy and various models of physical systems 1968 edition

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