

2016
6.

Which one of the following sets of titres indicates a systematic error if the actual volume being measured is 85.2 mL?

- (a) 85.1 mL, 85.1 mL, 85.3 mL, 85.5 mL
- (b) 65.2 mL, 75.2 mL, 85.2 mL, 95.2 mL
- (c) 85.2 mL, 85.3 mL, 85.1 mL, 85.1 mL
- (d) 87.3 mL, 86.9 mL, 89.1 mL, 88.2 mL

2016
7.

In an acid-base titration, which of the following is *least* likely to cause an error in the calculated concentration?

- (a) using a funnel in the burette and leaving it in the same place for each titration
- (b) measuring the volume at the bottom of the meniscus
- (c) each member of the experimental team taking turns to measure the burette
- (d) rinsing the burette with distilled water before the titration

2016
8.

How many moles of a diprotic acid would be required to neutralise 1 mole of sodium hydroxide?

- (a) 0.5
- (b) 1.0
- (c) 1.5
- (d) 2.0

2016
10.

Which one of the following represents a conjugate acid-base pair?

- (a) N^3/CN^-
- (b) $\text{NH}_3/\text{NH}_4^+$
- (c) $\text{CH}_3\text{CH}_2\text{OH}/\text{CH}_3\text{CHO}$
- (d) $\text{H}_3\text{PO}_4/\text{PO}_4^{3-}$

2016
11.

Which of the following equations *best* represents the self-ionisation of water according to the Brønsted-Lowry model?

- (a) $\text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}^+(\text{aq}) + \text{OH}^-(\text{aq})$
- (b) $\text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{OH}^-(\text{aq})$
- (c) $2 \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + 2 \text{OH}^-(\text{aq})$
- (d) $2 \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{OH}^-(\text{aq})$

CHEMISTRY

6

2016
12.

The *best* definition of the equivalence point in an acid-base titration is the point at which the

- (a) indicator changes colour.
- (b) volume of acid equals the volume of base.
- (c) pH of the solution is 7.
- (d) mole ratio of acid to base is equal to their stoichiometric ratio.

15 Wace Mathematics Questions

Wenbin Ji



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