M9 N. 05-X X (X = 0.01, 0.03) MO (X=0.01 or , Ni: 0.04 49 = 0.5 x 0.95 = 0-475 M (mg nitrate) 0.475 = n = 0.05 => n = 0.02375 mol .. W= 6.0895 9) mynitrate NI -0.5 x 0.04 = 0.02 KM 0.02 = n = 0.001 = mol M: 0-2908 3) Cr 2 0.5 & 0.01 20-00B 0-005 - M = 0.00025 * 400.14 W=0.10035 9) cr altrate (2) × 20.03 Cr, Ni20.02 Mg W= 6-0895 of Ni = 0-5 + 0:02 20:01 M : n = 0.0005 me (n; (~= 0-14549) wi nitate Cr= 0.5 40.05 = 0.015 M 1 = 0.00075 mol w= 0.300105 37 Crainer 73

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Great Britain Department of Science and Art

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