

Analytical Chemistry
3rd Long Exam

PART I. MULTIPLE CHOICE. Choose the letter that corresponds to the **BEST** answer. Write [] when the answer is **NOT** in the choices. (23 pts)

1. Transfer of electrons from one reactant to another

- a. precipitation rxn
b. half rxn
c. redox rxn
d. neutralization rxn

2. What is the mole ratio of $\text{Ce}(\text{IV})$ to NO_2^- for the oxidation of nitrite to nitrate using Ce^{4+} in the presence of an acid:



- a. 1:2
b. 2:1
c. 1:1
d. 2:2

3. How many mole(s) of electron is(are) produced in the oxidation of sodium bismuthate (NaBiO_3) to BiO^+ and H_3O^+ under acidic conditions?

- a. 1
b. 2
c. 3
d. 4

4. Azide (N_3^-) was added in the determination of the dissolved oxygen in water sample to remove interferences caused by the presence of what ion?

- a. chloride
b. fluoride
c. nitrite
d. nitrate

5. Which of the following statements is(are) true?

- I. Prepared KMnO_4 solutions should be stored in amber bottles
II. KMnO_4 solutions are boiled to remove CO_2 which makes the solution basic
III. $\text{Mn}(\text{II})$ is the product of the degradation of KMnO_4
IV. KMnO_4 solutions are filtered to remove manganese(IV) oxide

- a. I only
b. I and II
c. I and IV
d. I, II and III

6. What is the oxidation state of carbon in oxalic acid?

- a. +2
b. +4
c. +6
d. +7

7. The complex formed between $\text{Cu}(\text{I})$ and 1,10-phenanthroline has a molar absorptivity of $7000 \text{ L cm}^{-1} \text{ mol}^{-1}$ at 435 nm, the wavelength of maximum absorption. What is the absorbance of a $5.17 \times 10^{-4} \text{ M}$ solution of the complex when measured in a 1.00-cm cell at 435 nm?

- a. 0.0431
b. 0.0432
c. 0.432
d. 0.431

8. In the presence of iodine, thiosulfate ion is oxidized quantitatively to tetrathionate ion. What is the diatomic formula of tetrathionate ion?

- a. $\text{S}_2\text{O}_7^{2-}$
b. $\text{S}_4\text{O}_7^{2-}$
c. $\text{S}_2\text{O}_3^{2-}$
d. $\text{S}_4\text{O}_6^{2-}$

9. In Winkler titration, the flocc is produced from

- a. oxidation of I⁻ to I₂
b. oxidation of $\text{Mn}(\text{II})$ to $\text{Mn}(\text{III})$
c. reduction of I to I₂
d. reduction of $\text{Mn}(\text{II})$ to $\text{Mn}(\text{III})$

10. A component of a UV-Vis spectrophotometer that restricts the radiation being measured to a narrow band that is absorbed or emitted by the analyte.

- a. light source
b. detector
c. radiation filter
d. wavelength selector

11. Component of a UV-Vis instrument that operates by blocking or absorbing all but a restricted band of radiation.

- a. light source
b. detector
c. radiation filter
d. wavelength selector

12. Oxidation of VO^{2+} using permanganate or cerium as standard solution produces $\text{V}(\text{OH})_5^{2+}$, what is the oxidation state of vanadium in $\text{V}(\text{OH})_5^{2+}$?

Analytical Chemistry Exam Multiple Choice

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Four the author became the first graduate student at the University of the Chinese Academy of Sciences in 1978 studying rare earth element geochemistry under the guidance of Professor Chengji Guo a renowned mineralogist and geologist specializing in Rare and Rare Earth Elements and an Academician of the Chinese Academy of Sciences He achieved remarkable success earning national recognition as a leading expert and receiving special government allowances of the State Council His career included a fruitful international collaboration at the Geological Survey of Canada where his research was highly praised Despite his achievements author faced political suppression which ultimately led him to settle in Canada His memoir vividly portrays his resilience dedication and optimism while offering insights into rural Chinese life fieldwork in Xinjiang and scientific expeditions in the Okinawa Trough Blending personal narrative with popular science the book provides an engaging and thoughtprovoking account of a life shaped by history and driven by passion for learning

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