

- 6 Fig. 6.1 shows a quantity of gas in a cylinder fitted with a piston P.

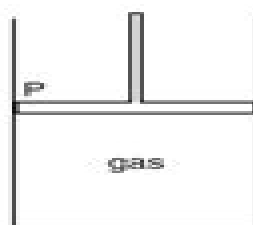


Fig. 6.1

- (a) Describe the motion of the molecules of the gas.

.....

 [3]

- (b) The piston is now slowly pushed down to decrease the volume of the gas. The temperature of the gas does not change.

- (i) State and explain, in terms of molecules, what happens to the pressure of the gas.

.....

 [2]

- (ii) Before pushing the piston down, the pressure of the gas was 1.0×10^5 Pa. Pushing the piston down reduces the volume of the gas from 500 cm^3 to 240 cm^3 .

Calculate the final pressure of the gas.

pressure = [2]

[Total: 7]

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