



Analytical Chemistry Interview Questions And Answers Guide

- Q1:** What is the difference between qualitative and quantitative analysis?
- A1:** Qualitative analysis is the determination of the presence or absence of a substance in a sample, while quantitative analysis is the determination of the amount or concentration of a substance in a sample.
- Q2:** What is the difference between a standard solution and a sample solution?
- A2:** A standard solution is a solution of known concentration, used for comparison with the sample solution. A sample solution is the solution of the substance being analyzed.
- Q3:** What is the difference between a primary standard and a secondary standard?
- A3:** A primary standard is a substance that is pure, stable, and has a known composition. A secondary standard is a substance that is not pure, stable, or has a known composition, but is used for comparison with the sample solution.
- Q4:** What is the difference between a titration and a gravimetric analysis?
- A4:** Titration is a method of quantitative analysis in which a solution of known concentration is added to a solution of unknown concentration until the reaction is complete. Gravimetric analysis is a method of quantitative analysis in which the mass of a substance is determined by weighing it.
- Q5:** What is the difference between a colorimetric method and a spectrophotometric method?
- A5:** A colorimetric method is a method of quantitative analysis in which the color of a solution is measured. A spectrophotometric method is a method of quantitative analysis in which the intensity of light passing through a solution is measured.
- Q6:** What is the difference between a chromatographic method and a mass spectrometric method?
- A6:** A chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their physical properties. A mass spectrometric method is a method of analysis in which the mass-to-charge ratio of ions is measured.
- Q7:** What is the difference between a gas chromatographic method and a liquid chromatographic method?
- A7:** A gas chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their volatility. A liquid chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their solubility.
- Q8:** What is the difference between a gas-liquid chromatographic method and a gas-solid chromatographic method?
- A8:** A gas-liquid chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their volatility and solubility. A gas-solid chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their volatility and adsorption.
- Q9:** What is the difference between a gas-liquid chromatographic method and a gas-solid chromatographic method?
- A9:** A gas-liquid chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their volatility and solubility. A gas-solid chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their volatility and adsorption.
- Q10:** What is the difference between a gas-liquid chromatographic method and a gas-solid chromatographic method?
- A10:** A gas-liquid chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their volatility and solubility. A gas-solid chromatographic method is a method of separation and analysis in which a mixture is separated into its components based on their volatility and adsorption.

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