

Analytical Chemistry  
3<sup>rd</sup> Long Exam

**PART I. MULTIPLE CHOICE.** Choose the letter that corresponds to the **BEST** answer. Write [ ] when the answer is **NOT** in the choices. (23 pts)

1. Transfer of electrons from one reactant to another

- a. precipitation rxn
- b. half rxn
- c. redox rxn
- d. neutralization rxn

2. What is the mole ratio of  $\text{Ce}(\text{IV})$  to  $\text{NO}_2^-$  for the oxidation of nitrite to nitrate using  $\text{Ce}^{4+}$  in the presence of an acid:



- a. 1:2
- b. 2:1
- c. 1:1
- d. 3:2

3. How many mole(s) of electron is(are) produced in the oxidation of sodium bismuthate ( $\text{NaBiO}_3$ ) to  $\text{BiO}^+$  and  $\text{H}_3\text{O}^+$  under acidic conditions?

- a. 1
- b. 2
- c. 3
- d. 4

4. Azide ( $\text{N}_3^-$ ) was added in the determination of the dissolved oxygen in water sample to remove interferences caused by the presence of what ion?

- a. chloride
- b. fluoride
- c. nitrite
- d. nitrate

5. Which of the following statements is(are) true?

- I. Prepared  $\text{KMnO}_4$  solutions should be stored in amber bottles
- II.  $\text{KMnO}_4$  solutions are boiled to remove  $\text{CO}_2$  which makes the solution basic
- III.  $\text{Mn}(\text{II})$  is the product of the degradation of  $\text{KMnO}_4$
- IV.  $\text{KMnO}_4$  solutions are filtered to remove manganese(IV) oxide

- a. I only
- b. I and II
- c. I and IV
- d. I, II and III

6. What is the oxidation state of carbon in oxalic acid?

- a. +2
- b. +4
- c. +6
- d. +7

7. The complex formed between  $\text{Cu}(\text{I})$  and 1,10-phenanthroline has a molar absorptivity of  $7000 \text{ L cm}^{-1} \text{ mol}^{-1}$  at 435 nm, the wavelength of maximum absorption. What is the absorbance of a  $5.17 \times 10^{-4} \text{ M}$  solution of the complex when measured in a 1.00-cm cell at 435 nm?

- a. 0.0431
- b. 0.0432
- c. 0.432
- d. 0.431

8. In the presence of iodine, thiosulfate ion is oxidized quantitatively to tetrathionate ion. What is the diatomic formula of tetrathionate ion?

- a.  $\text{S}_2\text{O}_7^{2-}$
- b.  $\text{S}_4\text{O}_7^{2-}$
- c.  $\text{S}_2\text{O}_3^{2-}$
- d.  $\text{S}_4\text{O}_6^{2-}$

9. In Winkler titration, the flocc is produced from

- a. oxidation of  $\text{I}^-$  to  $\text{I}_2$
- b. oxidation of  $\text{Mn}(\text{II})$  to  $\text{Mn}(\text{III})$
- c. reduction of  $\text{I}^-$  to  $\text{I}_2$
- d. reduction of  $\text{Mn}(\text{II})$  to  $\text{Mn}(\text{III})$

10. A component of a UV-Vis spectrophotometer that restricts the radiation being measured to a narrow band that is absorbed or emitted by the analyte.

- a. light source
- b. detector
- c. radiation filter
- d. wavelength selector

11. Component of a UV-Vis instrument that operates by blocking or absorbing all but a restricted band of radiation.

- a. light source
- b. detector
- c. radiation filter
- d. wavelength selector

12. Oxidation of  $\text{VO}^{2+}$  using permanganate or cerium as standard solution produces  $\text{V}(\text{OH})_5^{2+}$ , what is the oxidation state of vanadium in  $\text{V}(\text{OH})_5^{2+}$ ?

# Analytical Chemistry Exam Multiple Choice

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Researchers specializing in diversity issues were invited to help establish a dialog and spur discussion about how the results  
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In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Analytical Chemistry Exam Multiple Choice**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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